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LiCu₂O₂,

01.04.07

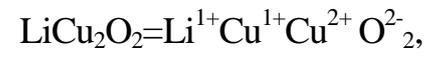
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- 2015

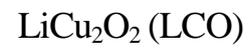
			5
1.			13
1.1.			
	-		13
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1.1.3.			18
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2.1.1.	-		45
2.1.2.			46
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2.3.			50
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2.3.2.			52
2.3.3.			53
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2.6.			57
2.7.			57
3.			60
3.1.			60
3.1.1.	-	LiCu ₂ O ₂	60
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3.4.1		LiCu_2O_2	83
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3.5.1.		LiCu_2O_2	88
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3.5.2.3	W2		100
3.5.3		LiCu_2O_2	103
3.5.3.1		$\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$	103
3.5.3.2		$\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$	105
			107
			110



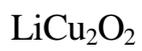
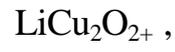
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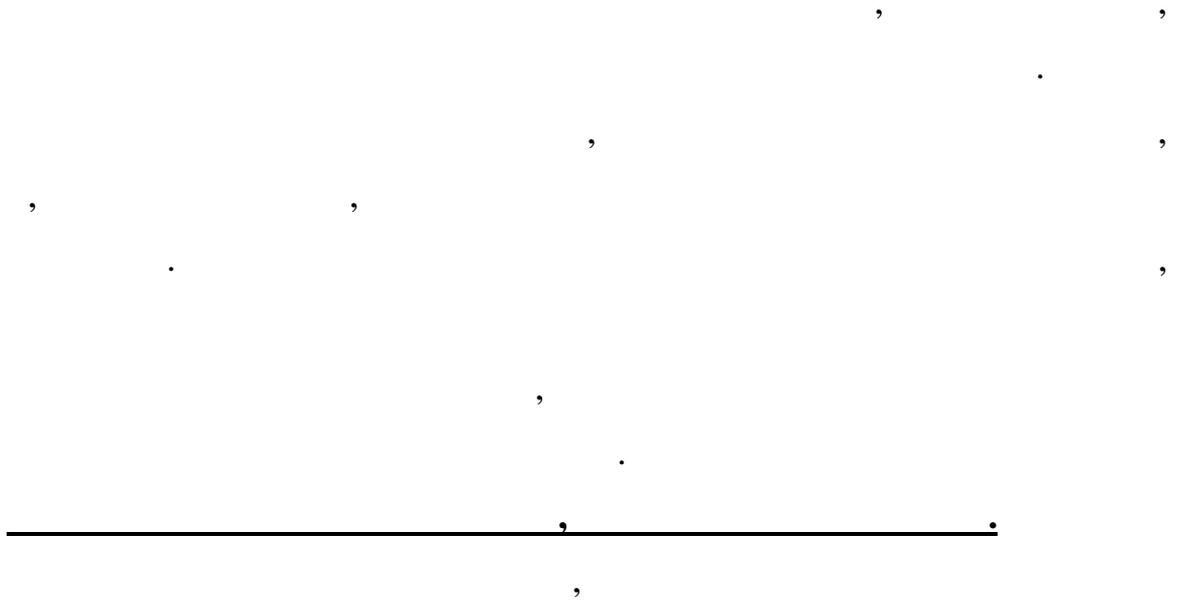


LCO

LCO,

123,





1. $\text{Li}_2\text{CuO}_2\text{-CuO}_x$;
 LiCu_2O_2 $\text{Li}(\text{Cu,Zn})_2\text{O}_2$,
 $(\text{Li,Ag})\text{Cu}_2\text{O}_2$;
 $4 \cdot 10^{10}$; LiCu_2O_2 .
 2. $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$, $(\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2$
 Zn Ag $= 0 - 0,12$ $= 0 - 0,04$,
 LiCu_2O_2 .
 3. $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$, $(\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2$.
 DC,
 $() = \frac{AC}{4,2 - 300}$ LiCu_2O_2 $\text{Li}(\text{Cu,Zn})_2\text{O}_2$,
 $(\text{Li,Ag})\text{Cu}_2\text{O}_2$ $0,1 - 100$.
 $3 \cdot \text{LiCu}_2\text{O}_2$
 $\text{Li}(\text{Cu,Zn})_2\text{O}_2$, $(\text{Li,Ag})\text{Cu}_2\text{O}_2$ DC $T \sim 300 \text{ K}$

$(DC = \sigma_0 \exp(E_a/k_B T))$

$$\sigma_{DC} = A \cdot \exp(T_0/T)^{1/4}$$

~25 DC

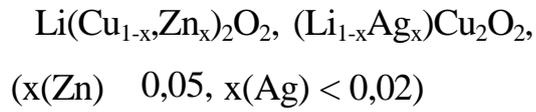
3 .



a, b c: a : b :

$$c = 2 : 1 : 10^4$$

3 .



S-

4.

LCO

LCO

O,

150 K.

O

(Li,Ag)Cu₂O₂ LiCu₂O₂ : Li(Cu,Zn)₂O₂,

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-

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LiCu₂O₂

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, : 4- . . «
(11), , 2011 ; XLVIII

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, , , -2012 ; IL

, , , , 2013 ; L

, , , , 2014 ; 63-

- «

. 12 - 26 2014 , , .

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... . . ().

_____ .
4 ,

, 5 .

_____ .

99

119 ,

59

5 .

_____ ,

CuO_4^-

(ladder compound),

LiCu_2O_2

_____ ,

_____ LiCu_2O_2

- ,
- ,
1. . . . DC AC
 $\text{LiCu}_2\text{O}_{2+}$ / . . . , . . . , . . . [.] //
 « . . . » . 2013. – 2. –
 .174–178.
 2. . . .
 , . . . , . . . , . . . , . . . , . . . , . . .
 // . 2015. .51. – 6. – .660–668 (Hieu
 Sy Dau. Effect of silver solubility on the structural, electrical, and magnetic
 properties of multiferroic LiCu_2O_2 / Hieu Sy Dau, K.E. Kamentsev, V.P.
 Sirotinkin, K.A. Yakovlev, E.A. Tishchenko, A.A. Bush // Inorganic Materials,
 2015. V. 51. – 6. – P. 598–606).
 3. . . .
 LiCu_2O_2 / . . . , . . . , . . .
 , . . . , . . . // . 2015.
 – 5. – . 716–720 (Sirotinkin V. P. X-Ray Diffraction Analysis of LiCu_2O_2 crystals
 with additives of silver atoms / V.P. Sirotinkin, A.A. Bush, K.E. Kamentsev, H.S. Dau,
 K.A. Yakovlev, and E.A. Tishchenko // Crystallography Reports. 2015. – Vol. 60. –
 5. – P. 662–666).
 4. . . .
 LiCu_2O_2
 / . . . , . . . , . . . , . . . //
 « . . . » . 2015. – 2. – .78–82.
 5. . . .
 LiCu_2O_2

- / . . . , . . . , . . . // .
4-
- « . . . » - « '11», .
3-7 2011 . N.
. - . 229-230.
6. . . DC AC
 $\text{LiCu}_2\text{O}_{2+}$ / . . . , . . . , . . .
. . . // . XLVIII
, . . . , . . . , 2012 , «
» - . 262-265.
7. . .
 $\text{LiCu}_2\text{O}_{2+}$ / . . .
, . . . , . . . , . . . // . IL
, . . . , . . . , 2013 .
« . . . » - . 188-191.
8. . . dc
 $\text{LiCu}_2\text{O}_{2+}$ / . . . , . . . , . . .
, . . . // . L
, . . . , . . . , -2014 . «
» - . 261-264.
9. . . , . . .
 $\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$ / . . . , . . . , . . . , . . .
, . . . // 63- -
« . . . , . . . , . . . , . . . , . . .
. 12 - 26 2014

1.

1.1

1.1.1

() ,

[1-3].

[4, 5],

[6-8],

[9-13],

[14]

()

(. . « »)

()

()

[15, 16].



[17].

« ».

(),

[15].

S-

[18, 19].

(

),

[20].

· ,

[15, 19, 20].

3d-

()

,
CuO₄,

SrCuO₂, Sr₂CuO₃,

SrCu₂O₃ Sr₁₄Cu₂₄O₄₁ [21–24].

[25].

[26, 27],

[24].

(. .) [28].

,
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 ,
 : , S-
 ,
 ,
 . . [22, 24, 29].

1.1.2

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 ()
 [30, 31].
 ,
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 ,
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 ,
 - .
 [30].
 . . . 1933
 ,
 (r_p) a .
 , $r_p \gg a$.

$$, r_p < a,$$

$$, r_p \sim a.$$

$$, - [32].$$

[9–13],

(n p)

(S=1/2),

[33].

()

(PA – phonon assisted). C

PA

(),

kT ,

[34 – 42].

() [30].

(N —

[31].

1.1.3

« » (,),

(. .)

μ
(DOS).

R .

« »

$\ln R \sim T^l$.

$$\ln R \sim T^n, \quad n < 1 \text{ [43–45].}$$

[43 – 45].

$$p_1(1-p_2)+p_2(1-p_1), \quad p_1 \quad p_2 -$$

(. [46] [47]).

:)

())

().

() E_a

$$(\max - \mu, \max + \mu) \text{ [44].}$$

[44]

i)

$$\tau = \tau_0 \exp\left(-\frac{V}{k_B T}\right), \quad (1.1)$$

ii)

v.

(1.1).

p

:

$$\tau = \tau_0 \exp(-V/k_B T),$$

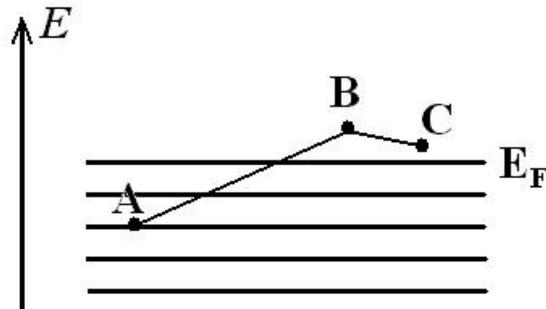
, $k_B T$

) ,) (

$$\exp(-2R),$$

$1/$

1).



1.1.

, A () B B C [44].

$$\dagger \approx \left(\frac{e^2}{k_B T} \right) p R^2 N(E_F) k_B T \approx e^2 p R^2 N(E_F), \quad (1.2)$$

$$p = \epsilon \exp\left(-2r R - \frac{v}{k_B T}\right), \quad (1.3)$$

$R -$, $N(E_F) -$. $N(E)$.

, :

$$v \approx \frac{1}{R^3 N(E)} \quad (1.4)$$

,

.

+ R [44]:

$$\left(\frac{4f}{3} \right) R^3 N(v) dv, \quad (1.5)$$

R

$$\Delta v = \frac{3}{4f R^3 N(E_F)}, \quad (1.6)$$

$$\epsilon \exp\left\{-2r R - \left[\left(\frac{4f}{3}\right) R^3 N(E_F) k_B T\right]^{-1}\right\}. \quad (1.7)$$

$$(1.7) \quad R_{opt}$$

$$2r = \left[\frac{9}{4} f R^4 N(E_F) k_B T \right]^{-1} \quad (1.8)$$

$$R_{opt} \quad (1.7) \quad (1.8),$$

$$\epsilon = \exp\left(-\frac{B}{T^{1/4}}\right), \quad B \approx 2,1 \left[\frac{\Gamma^3}{k_B N(E_F)} \right]^{1/4}, \quad (1.9)$$

$$(1.2) \quad (1.3) \quad ,$$

$$\dagger = \dagger_o \left(\frac{T_o}{T} \right)^{1/4}, \quad (1.10)$$

$$T_M \quad [36, 48]$$

$$T = 2,1^4 \cdot [\Gamma^3 / k_B N(E_F)] \quad (1.12)$$

:

$$N(E_F)$$

.

,

,

$$| - \mu | \quad ,$$

$$n() = 2N(E_F) \quad .$$

$$(1.8) \quad n()$$

$$R_{opt} [n()]^{-1/3} = [(9/2) \Gamma^3 N(E_F) k_B T]^{-1/4}, \quad (1.13)$$

opt

$$(1.13)$$

$$R_{opt} = [(9/2) \Gamma^3]^{3/4} ((k_B)^{3/4} [N(E_F) \Gamma^3]^{1/4}) = k_B^{-3/4} T_o^{1/4}, \quad (1.14)$$

$$, \quad (1.13) \quad (1.14)$$

$$(= \exp\{E_d/k_B T\}) \quad ,$$

$$() \quad , \quad ,$$

$$\max ()$$

$$\min () \quad .$$

$$\max / k_B = T_o^{1/4} \cdot T^{-3/4} \quad (1.15)$$

$$T \quad (\ln, T^{1/4}) \quad (\ln, T^{-1}),$$

$$\dots \quad \dots \quad [43] \text{ (SE)}, \quad \dots$$

max

max

$$\tau = \tau_0 \exp\left(-\frac{T_{SE}}{T}\right)^{1/2}, \quad (1.16)$$

$$T_{SE}$$

$$T_{SE} = \frac{s_{SE} e^2 r}{|k_B|}, \quad (1.17)$$

$$s_{SE} = 2,8,$$

$$|k_B| = (TT_{SE})^{1/2}.$$

$$T < T_V,$$

$$\tau_{opt}(T)$$

$$(T_V) = k_B(T_V T_{SE})^{1/2},$$

$$T_V,$$

$$(\ln, T^{-1/2}).$$

[37, 43, 44]

$$\tau = \tau_0 \exp\left(-\frac{T_0}{T}\right)^\epsilon, \quad (1.18)$$

$$\tau_0 = AT^{-m}, \quad (1.19)$$

A m -

$$= 1 - 1/4$$

$$(1.5), (1.7) \quad (1.8),$$

$$= 1/(d+1), \quad d -$$

$$m \quad (1.19)$$

$$[43] \quad 3 -$$

$$m_{opt} > , \quad [37] \quad SE \quad m_{opt} \cdot$$

$$= (m_{opt} a/2 s)^2,$$

$$, s = /q - ,$$

$$m_{opt} a/2 s \sim qa \quad qa > 1$$

$$F(r) \sim \exp(-r/a) \quad = 1/2 - m(\ll 1) = 1/2$$

$$m(\gg 1) = 9/2; \quad = 1/4 - m(\ll 1) = 1/4 \quad m(\gg 1) = 25/4.$$

$$F(r) \sim r^{-1} \exp(-$$

$$r/a) \quad = 1/2 - m(\ll 1) = -3/2 \quad m(\gg 1) = 5/2; \quad = 1/4 - m(\ll 1) =$$

$$- 3/4 \quad m(\gg 1) = 21/4 [37].$$

(SP).

W.

[48].

W.

$$w(\rho, W) = \exp\{-2r/a - (\rho + W/2)/T\}, \quad (1.20)$$

$$r = N^{1/d} = (G W)^{-1/d}, \quad (1.21)$$

$N - G$ (DOS), $d = 2, 3$

$$G(\rho, W_0).d .dW - W (W_0, W_0 + dW). \quad G \quad (1.21)$$

(1.20), $w(\rho, W) = W$

$$W_{opt} = W_{opt}/2 = T[T_o^{(d)}/T]^{2/(d+2)}, r_{opt} = (a/2)[T_o^{(d)}/T]^{2/(d+2)}, \quad (1.22)$$

$$T_o^{(d)} = [t_o^{(d)}/Ga^d]^{1/2}, \quad (1.23)$$

$t_o^{(d)} = 21,1 \quad 31,2 \quad d = 2 \quad 3,$

SP

(1.18) (1.19), $= 2/(d+2)$:

$$= \rho \exp\{-[T_o^{(d)}/T]^{2/(d+2)}\} \quad (1.24)$$

(W_{min} ,

$W_{max}), > T_1^{(d)}$

$W_{opt}(\rho) > W_{max} = W_{opt}(\rho_1^{(d)}),$:

$$t_i^{(d)} = \{[t_o^{(d)}W_{max}/4]^{d+1}/T_M^{(d)}\}^{1/d}, T_M^{(d)} = M^{(d)}/ga^d \quad (1.25)$$

$t_o^{(d)} = 0,546 \quad 0,607 \quad M^{(d)} = 13,8 \quad 17,0$

$d = 2 \quad 3,$, $g -$ DOS.

SP

()

:

$$= \exp\{-[T_M^{(d)}/T]^{1/(d+1)} - dW_{max}/T\}, \quad (1.26)$$

$$d = 0,189 \quad 0,174 \quad d = 2$$

3.

(1.26),

,

g,

G

-

.

$$= [(e^2/k)^d g]^{1/(d-1)} = [(e^2/ka)^d T_M^{(d)}]^{1/(d-1)}, \quad (1.27)$$

SE

$$< T_2^{(d)},$$

$$>_{opt} (T_2^{(d)})$$

$$T_2^{(d)} = [T_o^{(d)}]^{2/d}, \quad (1.28)$$

$$[T_2^{(d)} < T < T_1^{(d)}],$$

,

,

.

,

$$W_{min}, \quad (1.24)$$

$$W_{opt}(T_3^{(d)}) < W_{min}, \quad T_3^{(d)} = (W_{min}/W_{max})^{(d+1)/d} T_1^{(d)}.$$

(1.24)

$$\max[T_2^{(d)}, T_3^{(d)}] < T < T_1^{(d)},$$

$$(W_{max} - W_{min})/W_{min} \gg 1.$$

$$[2(T^d T_M^{(d)})]/W_{min} \ll 1 \ll T_3^{(d)},$$

:

$$= \exp\{-W_{min}/2T - [s_d T_M^{(d)}/T]^{1/(d+1)}\}, \quad (1.29)$$

$$s_d = 0,643$$

$$d = 2$$

$$0,63$$

$$d = 3.$$

$$W_{max} - W_{min}$$

opt,

,

(1.24)

$$W_{max} - W_{min}$$

$$W_{max} > \dots, \quad (1.26),$$

(1.29).

(1.18) (1.19).

$$m,$$

$$()$$

$$E_D = -d \ln / d(1/k_B T)$$

(1.18) (1.19) :

$$\ln[(E_D/k_B T) + m] = \ln + \ln T_{oj} + \ln(1/T), \quad (1.30)$$

$$m \quad (1.30)$$

$$\ln(1/T),$$

$$T_{oj} (, T_{SE} T_o^{(d)})$$

$$m, T_{oj},$$

(1.18) (1.19),

$$= 1 \quad m = 1 (\quad) \quad m = 3/2$$

().

[44].

LSMFO [47, 50])

m

LCMFO [37]

(),
 [37].

[37, 47], $m = 25/4$

9/2

- .
 ,

,

(DOS)

.

DOS

.

,

. 1.2

.

,

[45].

,

SP, DOS

$g(E_F) \sim 0,$

,

.

$W_{\min}/2.$

,

,

$2E_d,$

$= W_{\min}/2 - E_d/2,$

(1.32)



1.2.

[43].

() < .

1 2

1

2. ,

1 = 2 2 =

1 ,

(. .)

[50, 51].

(d.c.),

()

$$() \sim C^s \tag{1.31}$$

s 0,8 [43, 44].

C

[44]

$$W_1 - W_2 \quad (W = W_1 - W_2),$$

W

n

D,

F

$$\frac{nDF \cos^2 \theta}{k_B T (1 + \tilde{S}^2 \dagger^2)} \left\{ 1 + \exp\left(\frac{\Delta W}{k_B T}\right) \right\}^{-1} \tag{1.32}$$

$F = \cos^2$, $1/3$.

$$\dagger(\check{S}) = \frac{nD^2\check{S}^2\dagger}{3k_B T} (1 + \check{S}^2\dagger^2)^{-1} \left\{ 1 + \exp\left(\frac{\Delta W}{k_B T}\right) \right\}^{-1} \quad (1.33)$$

$W = 0$ $Nd(W)$ W $d(W)$,

W :

$$\int N \left\{ 1 + \exp\left(\frac{\Delta W}{k_B T}\right) \right\}^{-1} d(\Delta W) \quad (1.34)$$

N $NkT \ln 2$:

$$\dagger(\check{S}) = \frac{(\ln 2)ND^2\check{S}^2\dagger}{3(1 + \check{S}^2\dagger^2)} \quad (1.35)$$

[44] ,

U , :

$$\dagger(\check{S}) = \frac{f}{6} (\ln 2) NBk_B T \check{S} D^2 \quad (1.36)$$

$B -$.

, R ,

:

$$\dagger(\check{S}) = A \frac{e^2}{r^3} \{N(E_F)\}^2 k_B T \check{S} \left\{ \ln\left(\frac{\epsilon_{ph}}{\check{S}}\right) \right\}^4, \quad (1.37)$$

$A = (\pi^2/24)\ln 2 \approx 0,3$.

$() \sim s^{-1}$

,

.

, ph $10^{12} s^{-1}$,

$[\ln(\epsilon_{ph}/\check{S})]^4$ $-0,2$ 10^4

, $s \approx 0,8 -$

$$s = 2$$

[44].

Bottger Bryksin [46],

[43] .

A.R. Long

[51]

C

$$(1.19)$$

:

$$(\) \sim T^{n-s} \tag{1.38}$$

$s \ n$

()

()

tg

[35, 51 – 53].

$$\ddagger(T_{max}) \approx 1 \tag{1.39}$$

[44, 52, 54],

(= 1/2 f),

$$= \text{oiexp}(E_a/kT) \tag{1.40}$$

$f_o = 1/2 \ o$

. $E_a -$

tg

[52, 53] (tg)_{max} ~ n_0 .

1.2

LiCu₂O₂

1.2.1

LiCu₂O₂,

YBa₂Cu₃O_y [55–58].

[59]

LiCu₂O₂ [56]

Cu⁺

S=1/2 Cu²⁺

. . Pnma (. . 62, Z = 4),

a = 5,7286(2), *b* = 2,8588(1), *c* = 12,4143(3) Å

[55].

a/b

,

-

()

ab -

(.

1.3).

. 1.4

LiCu₂O₂,

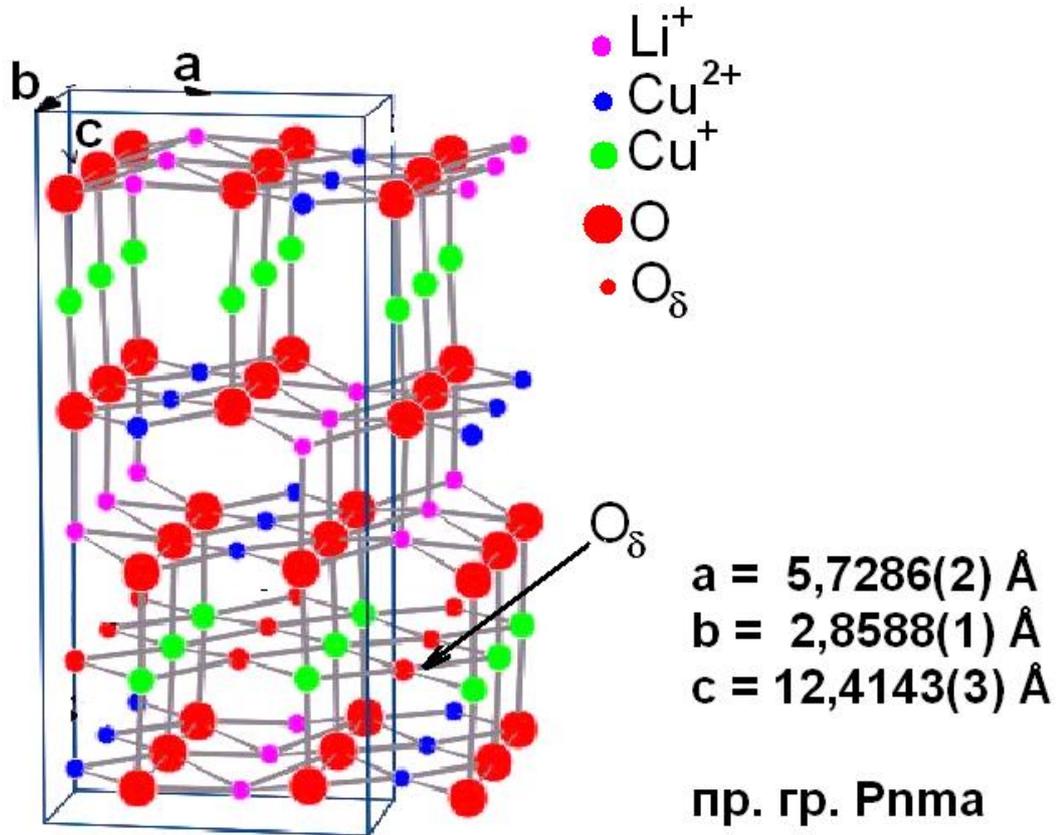
[55, 56, 58, 59].



1.3.

LiCu₂O₂,

[55].



1.4.

LiCu₂O₂

LiCu₂O₂

: 1) -Cu¹⁺(1)-, 2) -O(1)Cu²⁺(2)O(2)Li- 3) -

LiO(2)Cu²⁺(2)O(1).

Cu¹⁺

O²⁻-Cu¹⁺-O²⁻

LiCuO₂-

2) 3)

CuO₅ LiO₅,

ab-

LiO₄ CuO₄

a

Cu-O- Li-

O-

b -

Cu-O-

Li-O-

Cu-O-

CuO₅-

Cu-O (1,98 Å),

(2,48 Å),

Cu²⁺ [59].

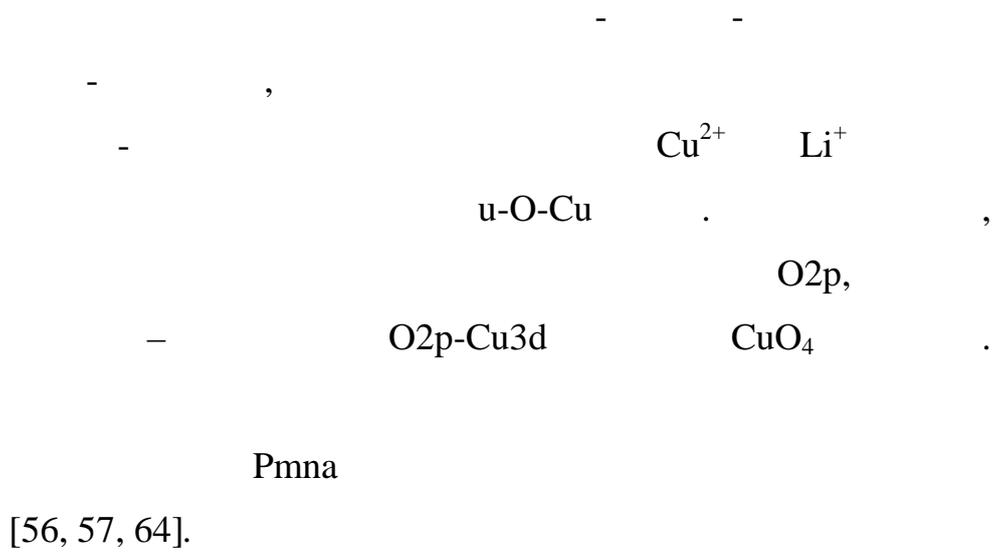
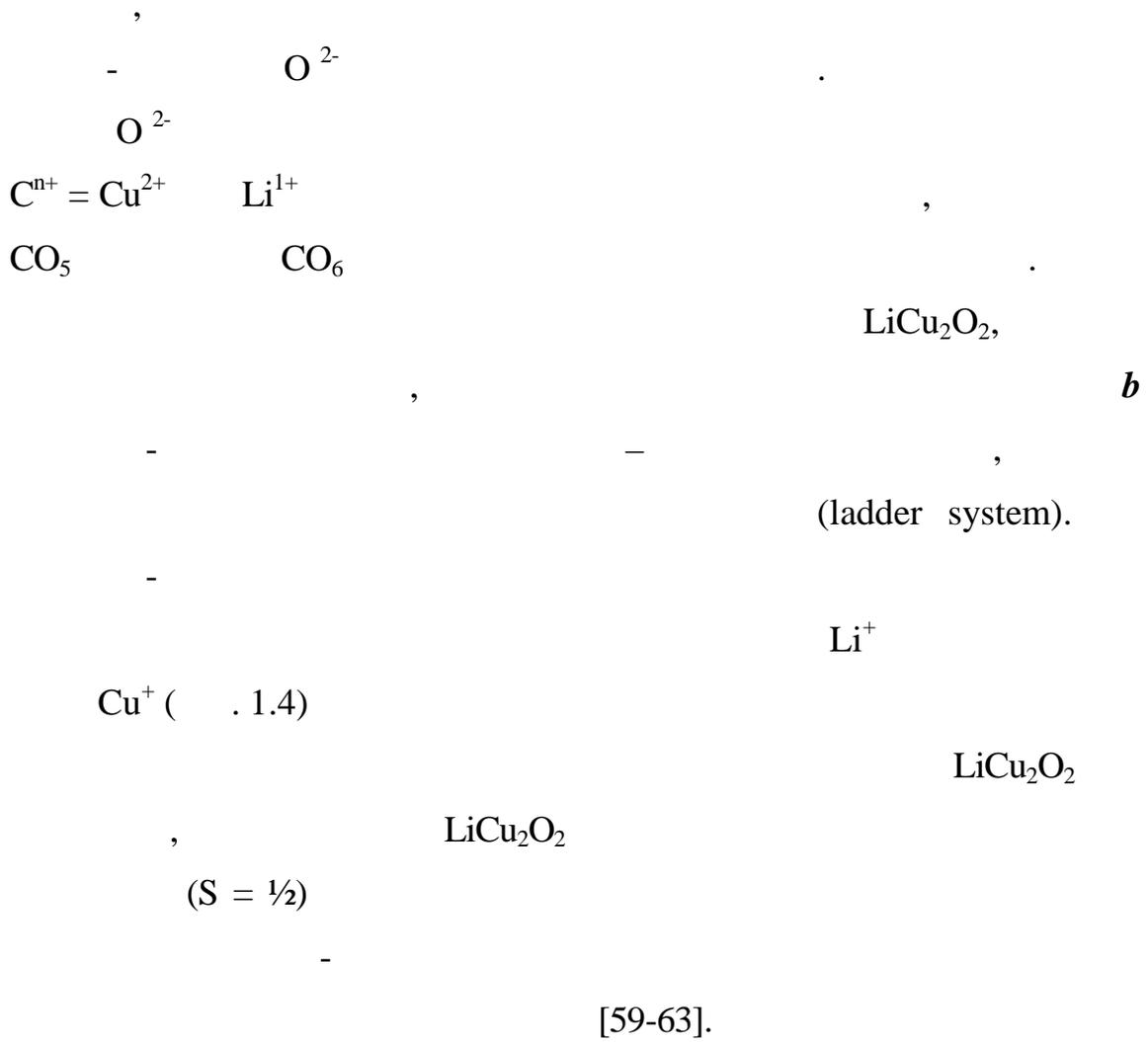
LiO₅

Li-O

d⁹

(~2,08 Å).

1) Cu¹⁺



1.2.2

LiCu₂O₂



b.



S=1/2 (two-leg ladder systems).



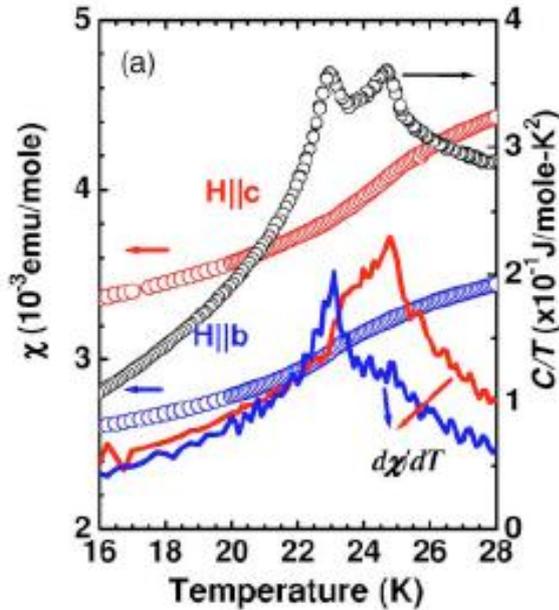
NN (*J*₁)

NNN

(*J*₂)

(*J*₃) (*J*₁ = -7,0, *J*₂ = 3,75, *J*₃ = 3,4

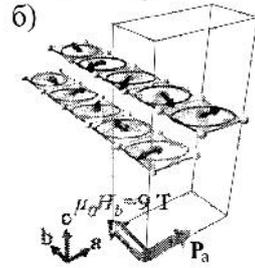
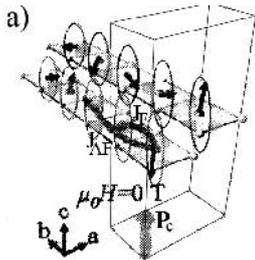
[66, 67]).



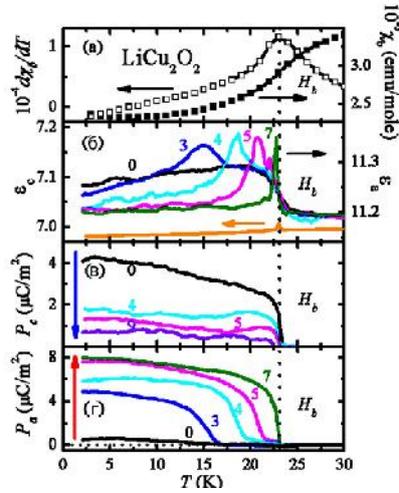
1.5.



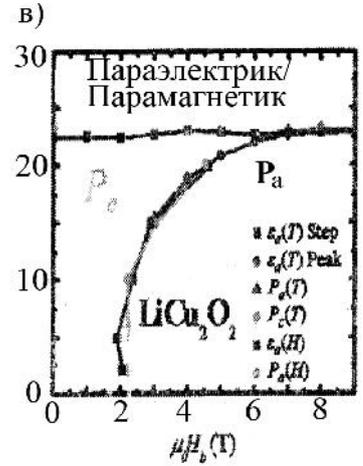
*T*_{c1}=24,6 *T*_{c2}=23,2 [68].



. 1.6.
1.6.



. 1.7.



. 1.8

Cu²⁺ Cu-O
()

H=9,0 T,
1.7. LiCu₂O₂:

b- () [65, 69].
: (a)

; ()
()

b

H_b=2

c a

H_b; ()

c

a

H_b (

) [65, 69].

1.8.

LiCu₂O₂,

P_c

P_a,

H_b.

Cu-O

T_{c1} = 24,6

T_{c2} = 23,2 (.1.5)

:

T_{c1} T_{c2}

T_{c2} [65–74].

[65, 68, 69, 72, 73, 75–78]

LiCu₂O₂

c

P_s.

T_{c2}

P_s

($\approx 1.6 - 1.8$). ,



II ().



($J \sim 10$),

(Cu-O-Cu) = 94 - 90 [56, 58].

1.2.3

LiCu₂O₂

LiCu₂O₂

[79, 80].

LiCu₂O₂

(\parallel)

()

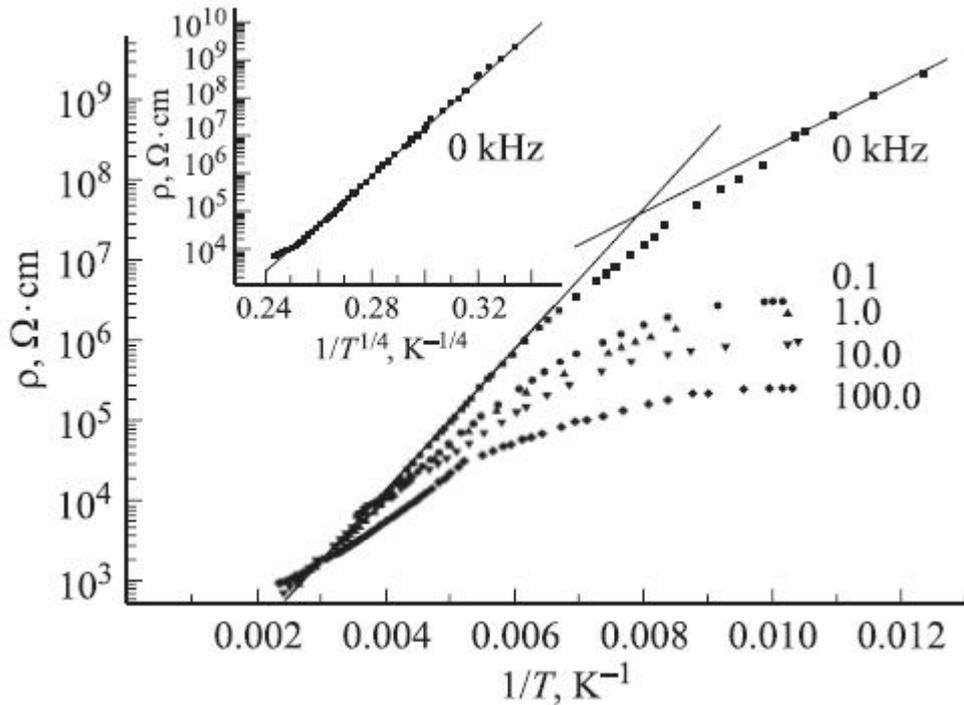
$\sim 10^3 \sim 10^2$ /cm

(.1.9).

LiCu₂O₂

$\lg -1/T$ (. 1.9).

$d(\lg \rho)/d(1/T)$



1.9

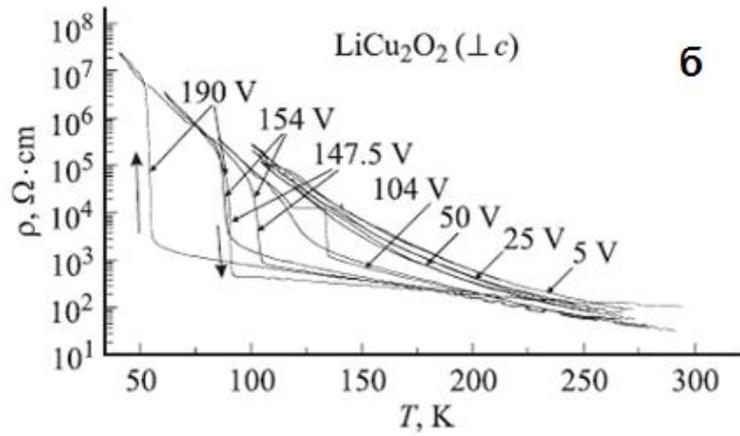
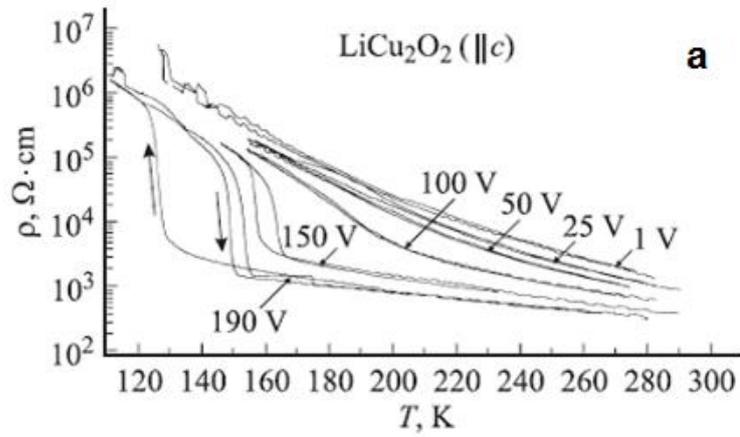
LiCu₂O₂

0.1, 1.0, 10.0, 100.0

$\lg -1/T$

(

$\lg -1/T^{1/4})/$



. 1.10

dc LiCu_2O_2 , ()
 () U ().

80–250

LiCu_2O_2

[79].

: () $\sim T^n$ (1.38).

s (1.38),

100 ,

$s \sim 0.2$

0.1–1.0

$s \sim 0.6$

10–100 .

260

,
 ,
 , [79].

LiCu₂O₂ U_o

(. 1.10). ,

LiCu₂O₂

,

,

 $U_o \approx 50$. $\sim 10^3$.

.

:

,

.

*ab.**c**ab* (. 1.10).

1.3

LiCu₂O₂

,

,

:

1)

LiCu₂O₂

II

,

,

/ , , ,
 . ,
 30) LiCu₂O₂ (~
 2) LiCu₂O₂
 S- -
 3) 123.
 4) ()
 ().

1.4

,

,

:

,

,

-

,

..

 LiCu_2O_2 .

,

.

,

.

,

.

.

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,

.

 LiCu_2O_2 ,

.

,

,

:

)

 LiCu_2O_2

;

) , - ;
)
(- - , , ,
) ;
) ;
) ,
.

2.

2.1

,
 ,
 .
 ,
 .
 -
 .
 ,
 ,
 .[81]

: 1)

- 2)

[81].

2.1.1 -

-
 .
 .
 ,
 « » (CuO – « . .», Li₂CO₃ « », ZnO « », AgNO₃
 « »).

,
 ,
 20÷100 50-100 .
 1 - 6 , 2 -
 50 / , -

() .

~200

2.1.2

-2-

() (.2.1) [82].

(5)

(5000).

V

- 20÷3000

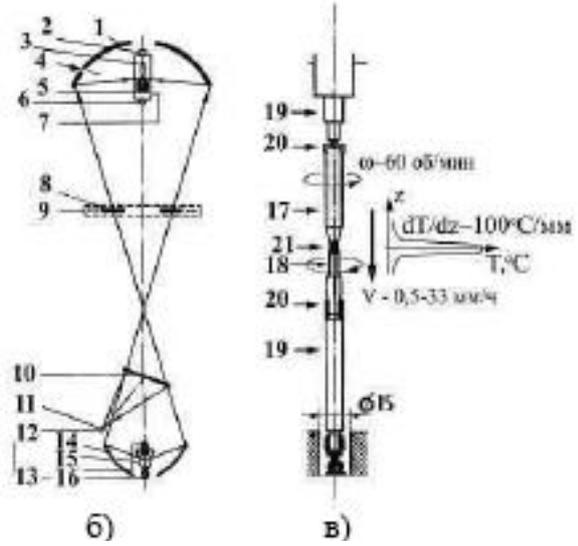
- d /dx ~ 100 / ;

- V = 0,5÷33 / .

= (10÷500) / .

- 12

10



а)

б)

в)

2.1.

()

(,)

-2- .1, 14 –

(), 15 –

, 3 –

, 11, 10 –

, 13, 16 12 –

,

, 8, 9 –

,

, 4 –

, 2, 6 –

, 17 –

, 18 –

, 19 –

, 20 –

, 21 –

.

(. 2.2),



.

([83–86]),

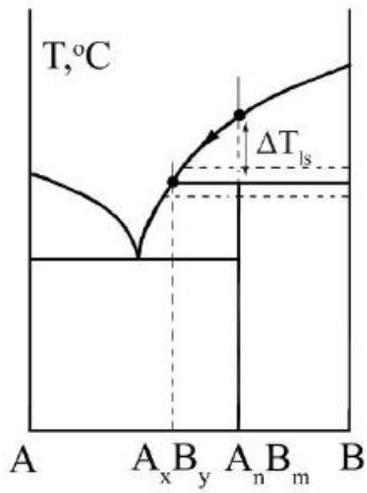
T_{ls}

D

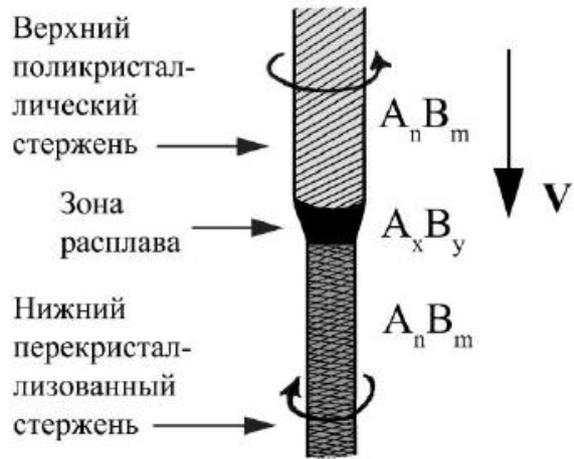
:

$$(d /dz)/V > T_{ls}/D$$

(3.1)



2.2.



$A_n B_m$
 $A_x B_y$
 $A_n B_m$
 V_c
 $V > V_c$
 V_c
 (dT/dz)
 $T_{ls} \sim 20$
 $V \ll 6$ / .
 90 4 8

~ 100 / ²,

3,0÷5,5 / ,

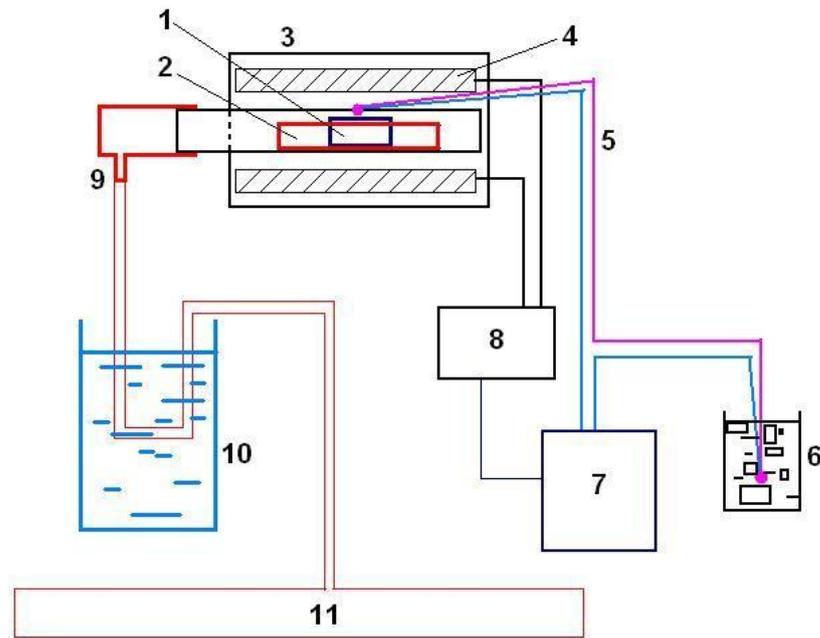
2.2

- O

LiCu₂O₂₊ ,

~ 840°C

()



2.3.

: 1 - , 2 -

, 3 - , 4 -

, 5 - Cu-(Cu-Fe)

6, 7 - () - 101,

8 - - 101 (), 9 -

, 10 -

, 11 - .

. 2.3.

,
- 101 (,) ± 1°C,

()

O

c

2.3

2.3.1

d

(

)

(

)

d

-

:

$$2d/\sin =$$

$$(3.2)$$

()

-3,

-4 (. 2.4)[87]

: (CuK₁) = 1,54051, (CuK₂) = 1,54433.

ICDD

[88].

2

. 2.5.

2÷5 /

- 0,25÷0,50 / .

Ge (a =

5,6567(6) Å

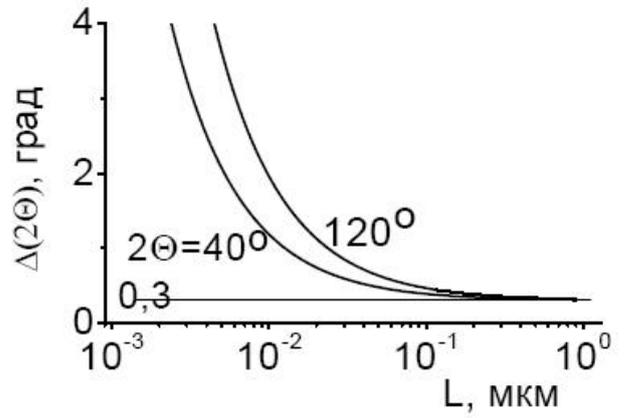
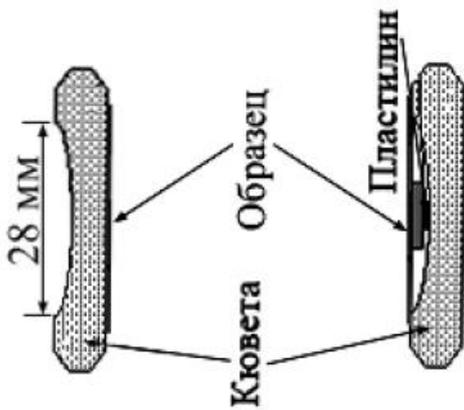
2

±0,02°.



. 2.4.

-4



2.5) () ()
 .)
 2

d

$$d/d = -ctg \quad (3.3)$$

2.3.2

a, b, c

:

$$2d/\sin = \quad (3.4)$$

$$1/d^2 = h^2/a^2 + k^2/b^2 + l^2/c^2 \quad (3.5)$$

h, k, l

CELREF [89].

2.3.3

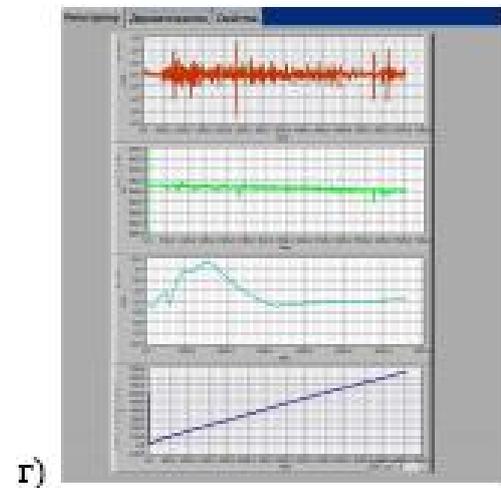
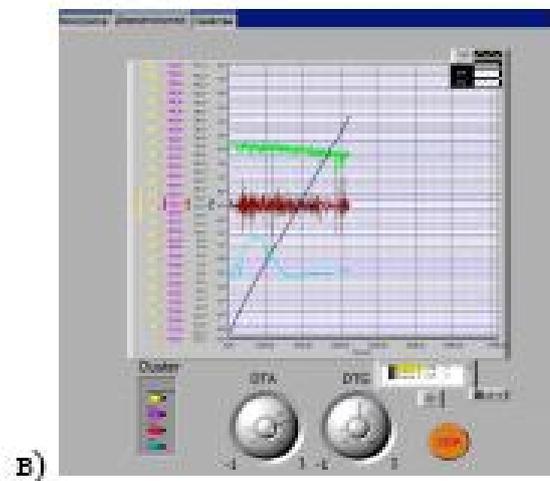
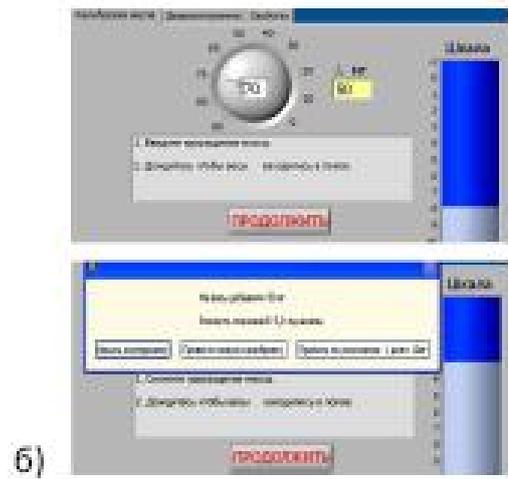
a, b c

- 2 .

2.4

()

()



2.6.

Q1500.vi

() .

,

.

Q1500 D

F. Paulik, J. Paulic, L. Erdey.

,

-

(),

,

,

.

,

.

2.5

,

-

.

2.7.

Au-Au:Fe,

0,1 ,

(.2.8)

-

.

5,26 .

-300

1

(10)

Lock_in

amplifier,

10^{-9} .

,

10^{-12} - 10^{-13} .

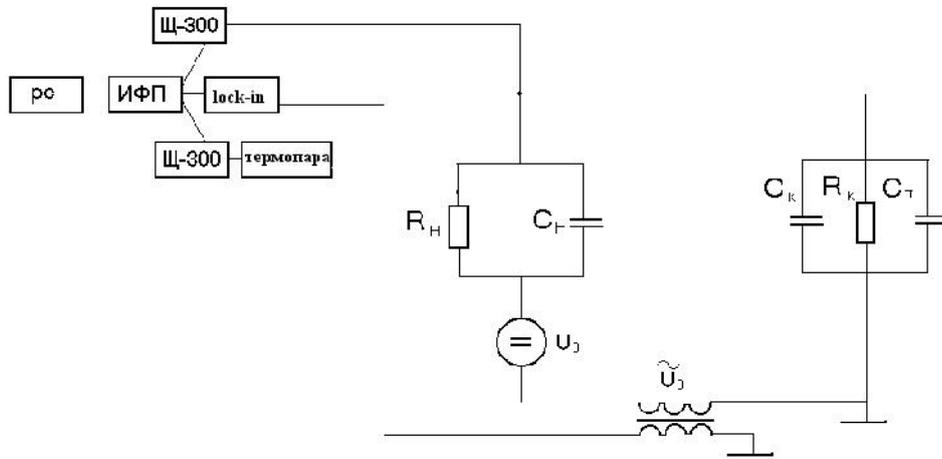
-

,

,

,

.



2.7.

, R C – , R C – , C –

U_o , $U_o \sim$, R

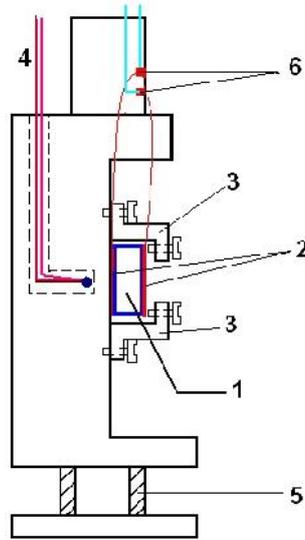
($0 < U_o < 250$)

($1.74 \cdot 10^3$, $U_o \sim = 100$)

($R = 5,26$)

() ... ()

Labview (2.9).



2.8.

In-Ga : 1 – , 2 –
 , 3 –
 , 4 – Au-Au:Fe , 5 – , 6 –

7-78/1 10 (. 2.9).

,
 ,
 7-78. 0^0

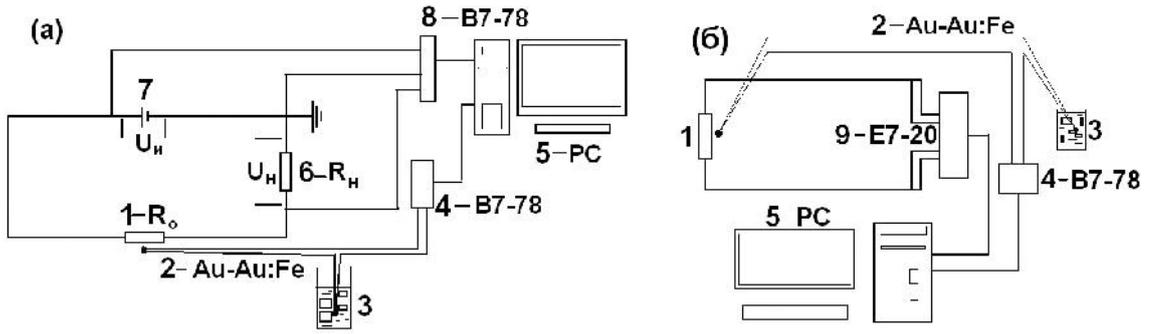
LabView.

$$R_o = \frac{U - U}{U} R$$

(3.6)

U R – , U –

Au-Au:Fe.



2.9.

() () : 1 - ; 2 - Au-Au:Fe
 3; 4 8 - 7-78; 5 -
 ; 6 - ; 7 -
 , 9 - E7-20.

$$\tau = \frac{l}{R_0 \cdot S} \quad (3.7)$$

- , l S - .

7-20

0,1-100 .

. 2.9 . ,
 , 7-78.

LabView.

2.6

- 300 SQUID MPMS-XL-7 Quantum
 Design Inc (H = 20) ZFC (zero- field-cooled
 - -) FC (field-cooled -).

2.7

:

() .

“Orbis” “EDAX” ()

Si(Li)

Na.

30

40

60

50

():

-2 (. 2.10)

-25

20-50

(25)



2.10. - " -2"

-5).

-2 10^{-4} - 10^{-5} %.

30 %.

-100,

3.

3.1

LiCu₂O₂
 (Li_{1-x}Ag_x)Cu₂O₂ Li(Cu_{1-x}Zn_x)₂O₂ 0 x 0,05 0 x
 0,12 .

3.1.1

LiCu₂O₂

[51-54]

LiCu₂O₂ 1163 – 1323 ,

. 3.1.

CuO

« . »

Li₂CO₃

« ».

LiCu₂O₂

xCuO·(1-x)Li₂CO₃ c

0,77

x

0,83

4

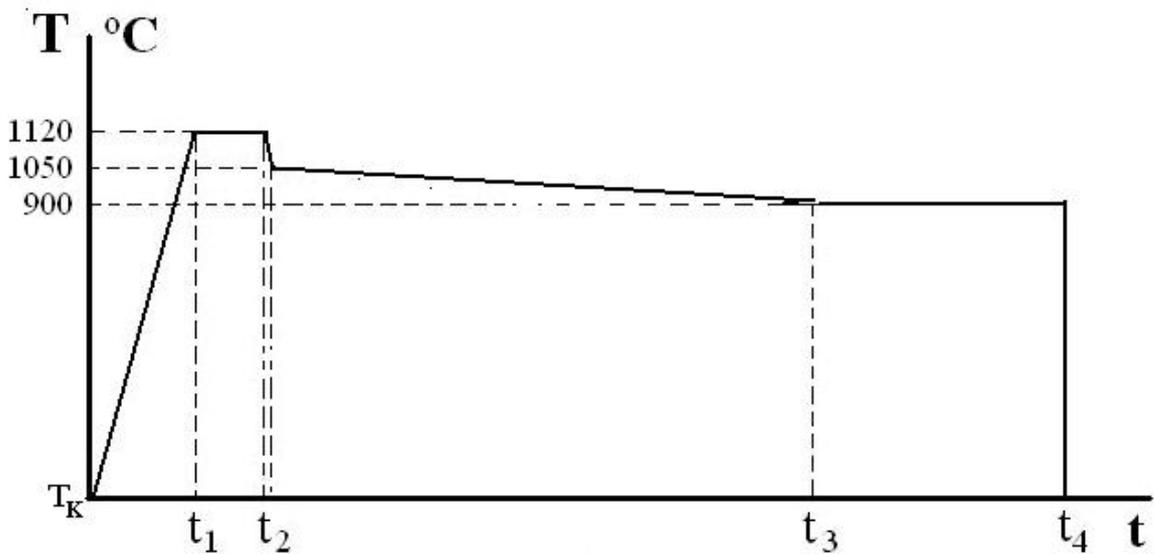
1393

0,5 ,

1323 ,

2,0 /

1173 ,



3.1

-

.



3.2 –

 LiCu_2O_2 .

1173 (20 - 24),

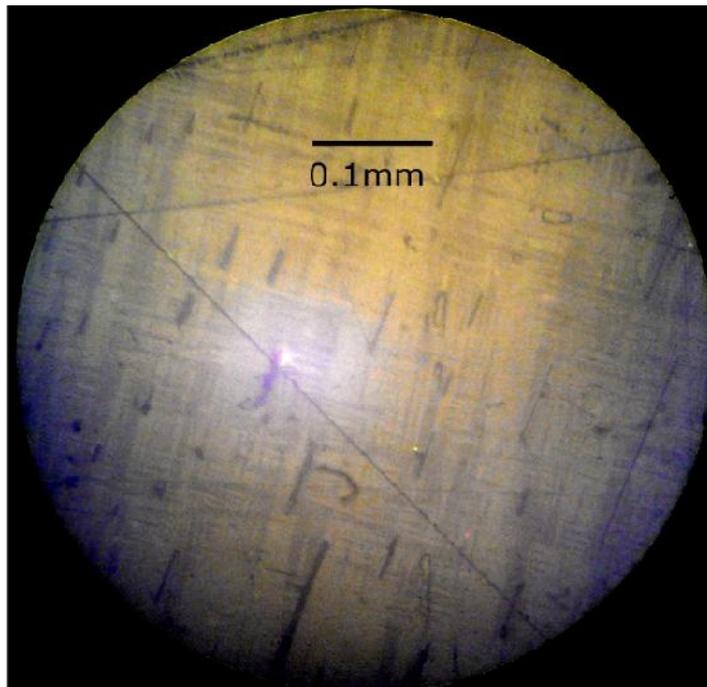
 LiCu_2O_2

1393

1323

~1173

 LiCu_2O_2 [. . 3.2.1]. LiCu_2O_2 (0,5–4) 8 8 ³ (. 3.2).



3.3.

 LiCu_2O_2

(001),

(210) (2-10),

{001}

{210}.

 LiCu_2O_2 ,

(120) (1-20).

(001)

(. 3.3).

5,2 / ³.

()

 LiCu_2O_2

3.

3.1.2.



Ag, Zn

CuO , AgNO₃ ZnO « » , « » , « » « » Li₂CO₃,
 Li₂CO₃·4(1-x)CuO·4xAgNO₃,
 Li₂CO₃·4(1-x)CuO 4xZnO - I (1-x)Li₂CO₃·2xZnO 4CuO – II 0 x 0,5,
 LiCu₂O₂.
 LCO,
 > 0,15
 1 .

3.1.3.



6

90 .



3.4.

-2-

()

5,0 / .

6

20 (. 3.4).

a

a

3.1.4.

LiCu₂O₂

LiCu₂O₂

~1113

40

LiCu ₂ O ₂				3.1.
1	-		2	400
2	-		-	
1			-	
2			-	
A4	-	40 .	2	4
		1113		400
N3	-	1	1	
W2	-	40 .	-	
		1113		

1113

: 400

1 - 4 .

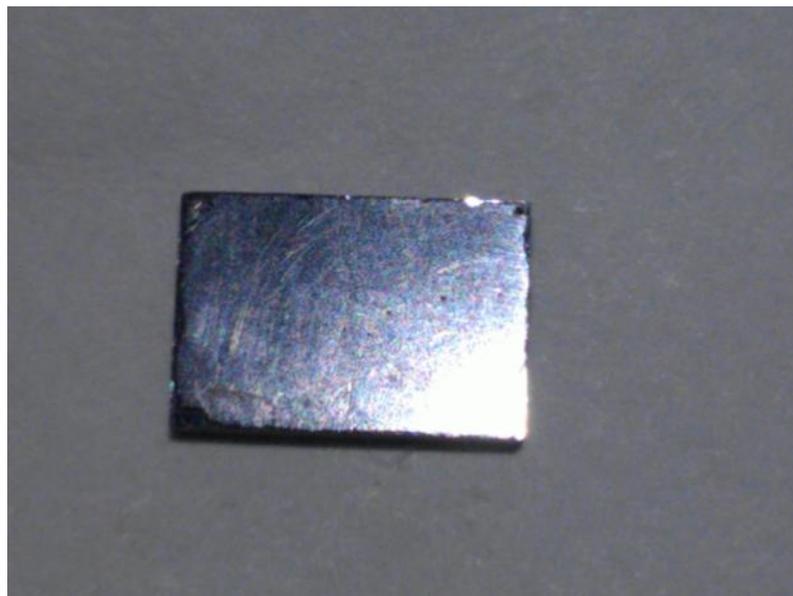
3.1



3.1.5.

(001), (210) (2-10) (.3.5).

In-Ga



3.5

3.2.

3.2.1

LiCu₂O₂:

Q-1500D

Pt-

 $T =$ 1200° - Al₂O₃.LiCu₂O₂

[90],

 $T = 553-773$ LiCu₂O₂ Li₂CuO₂ CuO.

(. 3.6)

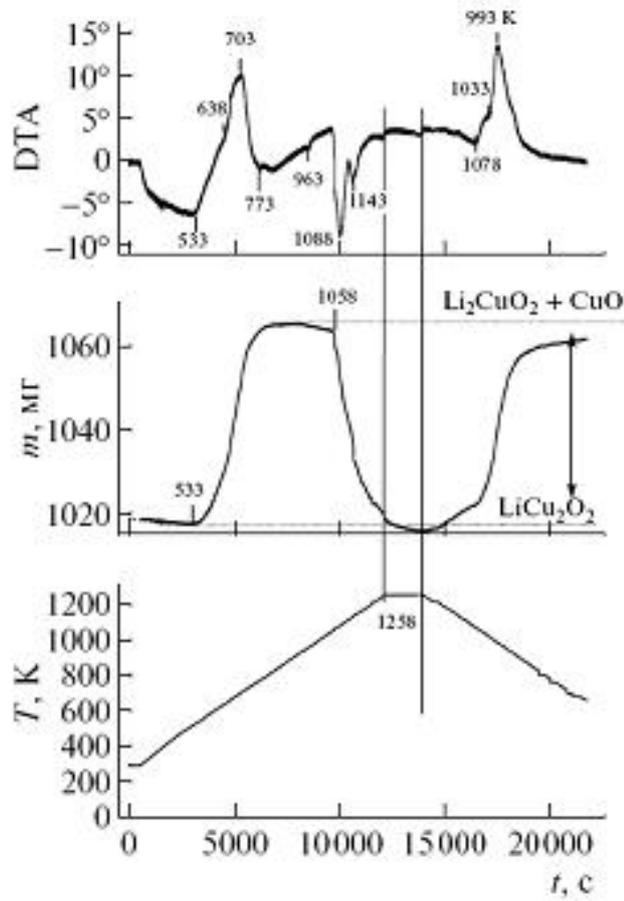
Cu⁺Cu²⁺,LiCu²⁺Cu⁺O₂LiCu₂O₂ (683 °) + O₂↓ → LiCu₃O₃ (783 °) + O₂↓ + Li₂CuO₂ → Li₂CuO₂ + CuO + O₂

1073-1223

LiCu₃O₃,LiCu₂O₂,

:

Li₂CuO₂ (1108 °) + O₂↑ + CuO → LiCu₃O₃ (1163 °) + O₂↑ + Li₂CuO₂ + → LiCu₂O₂LiCu₂O₂ (1163 ° < T < 1323 °),LiCu₂O₂



3.6.

LiCu_2O_2 ,

(DTA –

, m –

, T –

, t –

).

DTA

1163–1323

Li_2CuO_2

,

783 – 800

LiCu_2O_2

Li_2CuO_2

CuO .

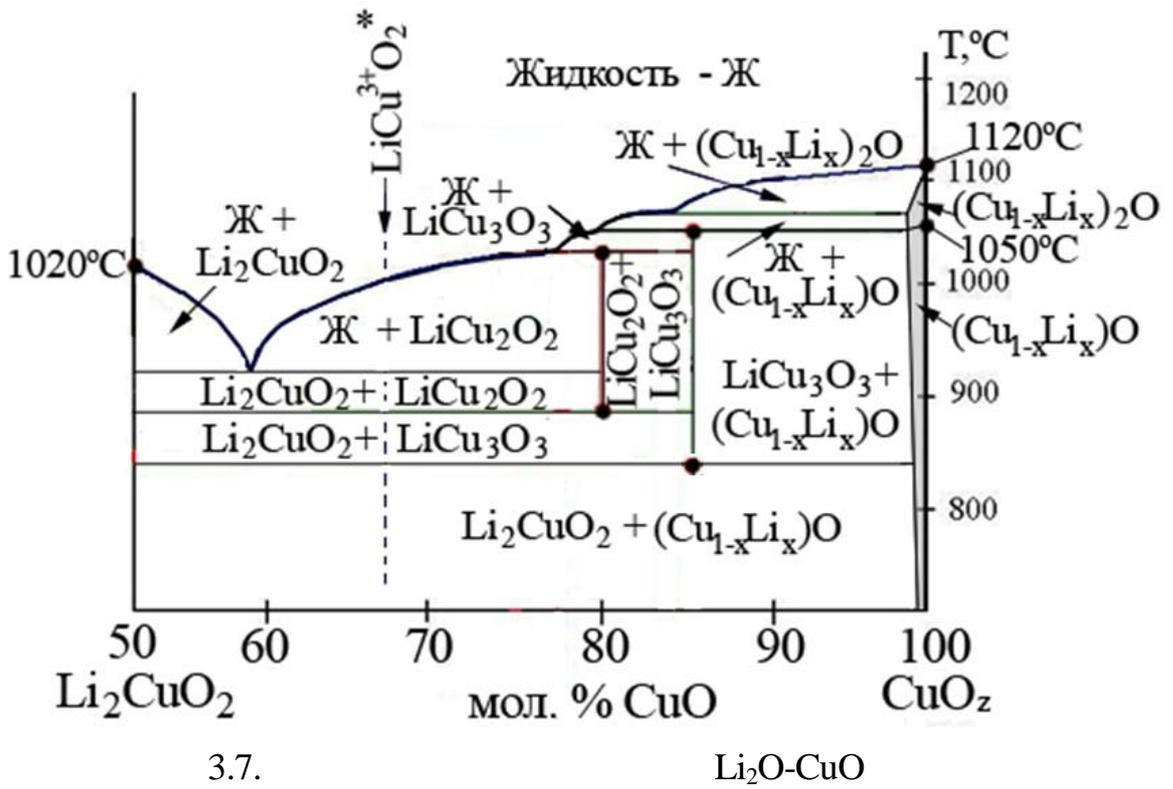
,

LiCu_2O_2

1163 – 1373 .

1323

LiCu_2O_2 .



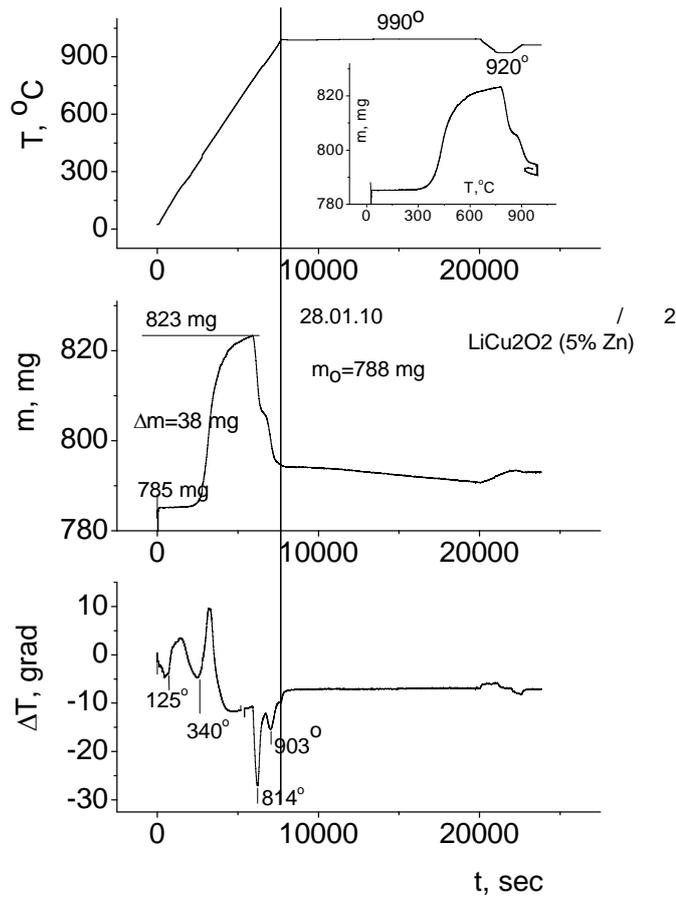
(. 3.7). LiCu_2O_2
 LiCu_3O_3 ,
 (1323 1373) (1163 1113
).

1173 - 1323

$\text{Li}(\text{Cu},\text{Zn})_2\text{O}_2$, 5 .% Zn

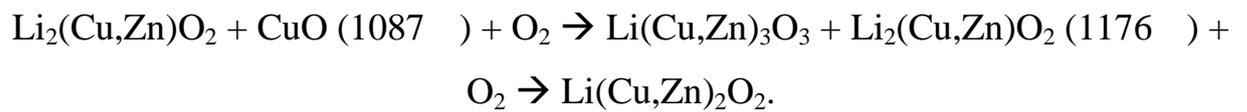
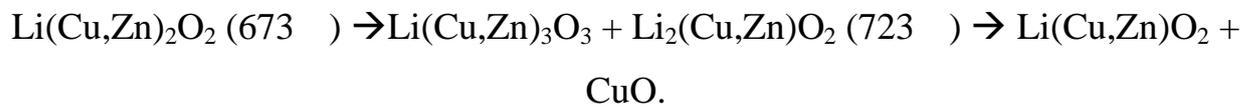
LiCu_2O_2 ; $\text{Li}(\text{Cu}_{0.95}\text{Zn}_{0.05})_2\text{O}_2$ (. 3.8)

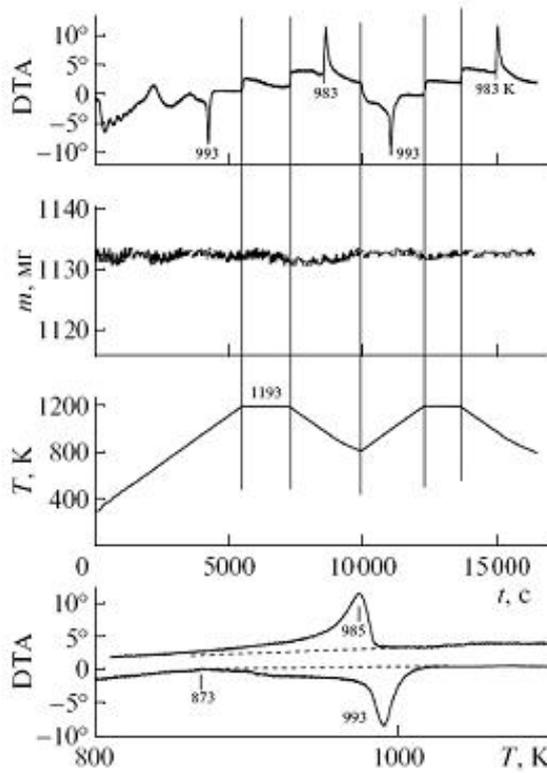
LiCu_2O_2 5 Zn



3.8.

Li(Zn_{0.05}Cu_{0.95})₂O₂: 7,5 / ,
788,0 , ()
1500 Al₂O₃) .





3.9.

LiCu_2O_2 ,

(DTA –

, m –

, T –

, t –).

DTA

3.2.2.

LiCu_2O_2

LiCu_2O_2

(1320).

$T_{PT} = 993$

(. 3.9).

DTA $T = 983$

DTA

3.3.

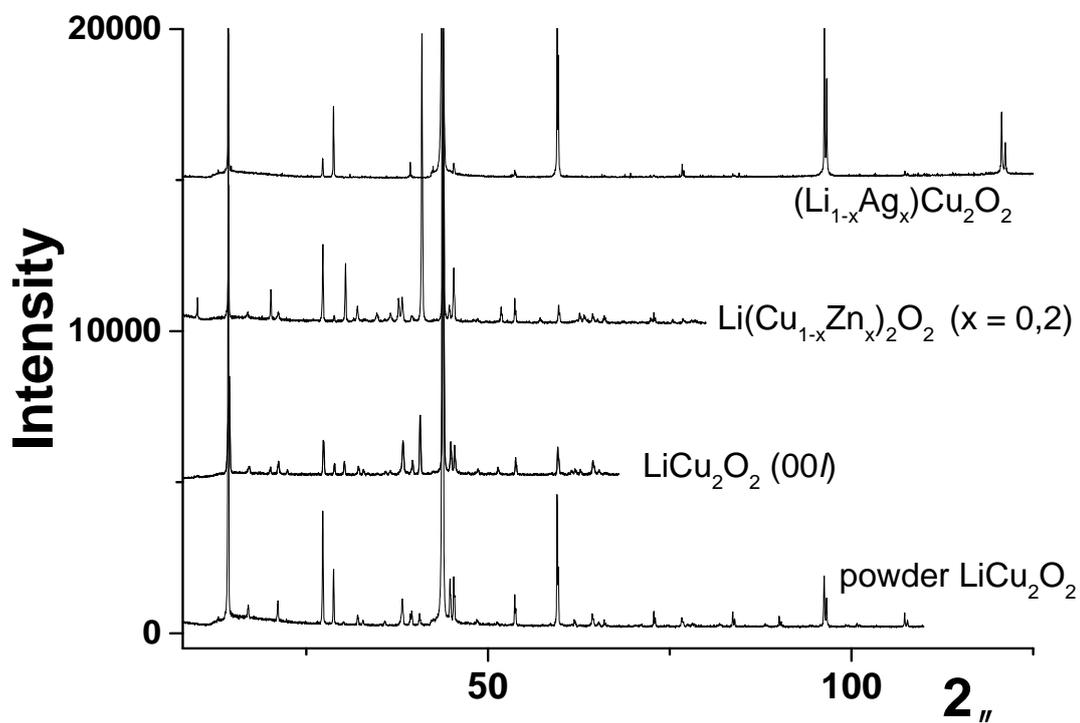
3.3.1



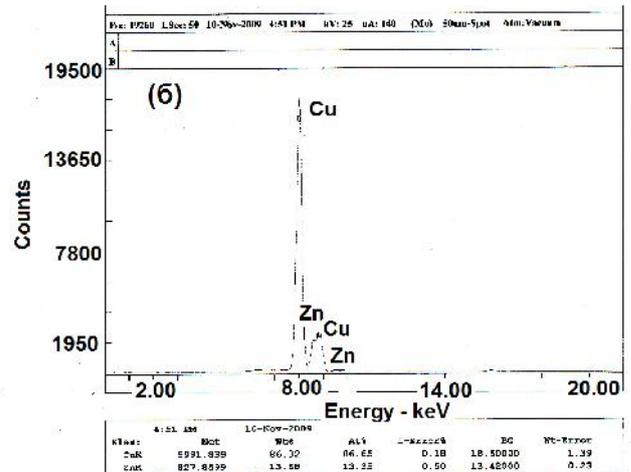
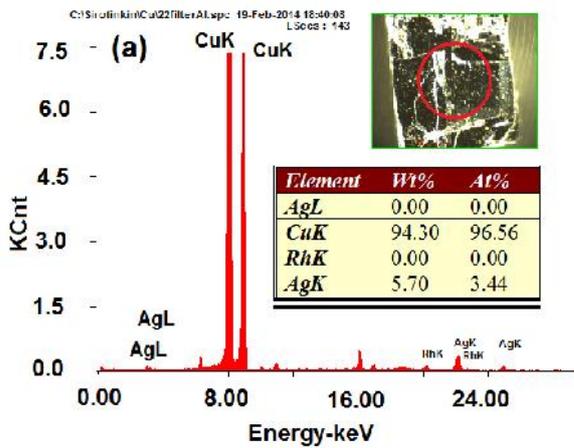
$$a = 5,726(2), \quad b = 2,858(1), \quad c = 12,410(2) \text{ \AA},$$

[52–54, 84] LiCu_2O_2 .

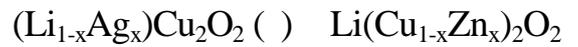
5%



. 3.10.



. 3.11.



().

(400) (020),

-2 (. 3.3.1).

ZnO

Zn



Zn

Li



3.3.2

3.3.2.1

Li O,

. 3.11.

Ag



4 .%

Ag x = 0,25.

$\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{ZnO}$ = 0,10 Zn , 12 .% () .

Li_2O_2 4 .% Ag 12 .% Zn.

$(\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2$ $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$ 0 x 0,05 0 x 0,12

3.3.2.2

–

3.2			
LiCu_2O_2 () .			
	%	%	% , .
Li	4,0552	18,8013	20
Cu	74,2941	37,6456	40
O	21,5616	43,3606	40
C	0,0605	0,1620	
Al	0,0129	0,0154	
Na	0,0032	0,0045	
Ca	0,0043	0,0034	
K	0,0017	0,0014	
Mg	0,0008	0,0011	
Si	0,0009	0,0010	
S	0,0010	0,0010	
Cl	0,0007	0,0007	
Fe	0,0010	0,0006	
Cr	0,0005	0,0003	
Mn	0,0005	0,0003	
B	0,0001	0,0002	
P	0,0002	0,0002	

			3.3
			LiCu ₂ O ₂ () .
	%	%	% .. .
Li	4,0552	18,8013	20
Cu	74,2941	37,6456	40
O	21,5616	43,3606	40
C	0,0605	0,1620	
Al	0,0129	0,0154	
Na	0,0032	0,0045	
Ca	0,0043	0,0034	
K	0,0017	0,0014	
Mg	0,0008	0,0011	
Si	0,0009	0,0010	
S	0,0010	0,0010	
Cl	0,0007	0,0007	
Fe	0,0010	0,0006	
Cr	0,0005	0,0003	
Mn	0,0005	0,0003	
B	0,0001	0,0002	
P	0,0002	0,0002	

- ()

, [Cu]/[Li]

2 - ,

[O]/[Li] 2,2 - 2,3 = 2 +

(. . 3.2 3.3).

= 0,2 - 0,3 O , ,

123 [91],

[92].

, O .

Al ,

•
•

3.3.3

Ag Zn

4 (CuK -).

[56]

LiCu₂O₂.

(204), (006), (210), (008),

(108), (400), (216), (200) (004), (006)

CELREF

a, b, c

(. 3.12).

Ag

a c, *b* $x = 0,25$. $x = 0,25$ LiCu₂O₂ 4

.% Ag

[93] (. . 3.4),

LuCu₂O₂Li⁺Cu²⁺ Ag²⁺.*c*(*c/c* $x \sim 20\%$)Ag²⁺ Li⁺ Cu²⁺ (. . 3.4).

. 3.4

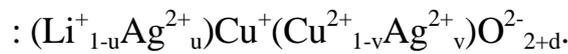
Li⁺, Cuⁿ⁺ Agⁿ⁺, n = 1, 2
(. .) [93]

3

= [r(Agⁿ⁺)-r(C^{m+})]/r(C^{m+}), n, m = 1, 2, 3 (. . =5
 . . = 4 . . =6, SQ - -)

	. . 2	. . 4	. . 5	. . 6	. . 8
Li ⁺ (1s ²)		0,590	0,675 *	0,76	0,92
Cu ⁺ (3d ¹⁰)	0,46	0,60	0,685 *	0,77	
Cu ²⁺ (3d ⁹)		0,57	0,65	0,73	
Cu ³⁺ (3d ⁸)				0,54	
Ag ⁺ (4d ¹⁰)	0,67	1,00 1,02 SQ	1,09	1,15	1,28
Ag ²⁺ (4d ⁹)		0,79 SQ	0,865 *	0,94	
Ag ³⁺ (4d ⁸)		0,67 SQ	0,705 *	0,74	
(Ag ⁺ -C ⁿ⁺) C ⁿ⁺ =Li ⁺		0,69	0,615	0,51	0,39
Cu ⁺	0,46	0,67	0,59	0,49	
Cu ²⁺		0,75	0,68	0,57	
Cu ³⁺				1,13	
(Ag ²⁺ -C ⁿ⁺) C ⁿ⁺ =Li ⁺		0,34	0,28	0,24	
Cu ⁺		0,32	0,26	0,22	
Cu ²⁺		0,39	0,33	0,29	
Cu ³⁺				0,74	
(Ag ³⁺ -C ⁿ⁺) C ⁿ⁺ =Li ⁺		0,135	0,04	-0,03	
Cu ⁺		0,12	0,03	-0,04	
Cu ²⁺		0,175	0,08	+0,01	

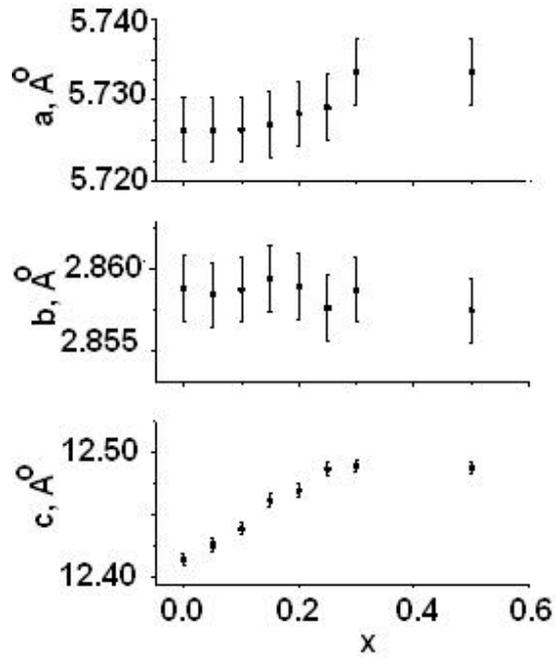
O²⁻ Ag²⁺ Li⁺, 1) Cu¹⁺



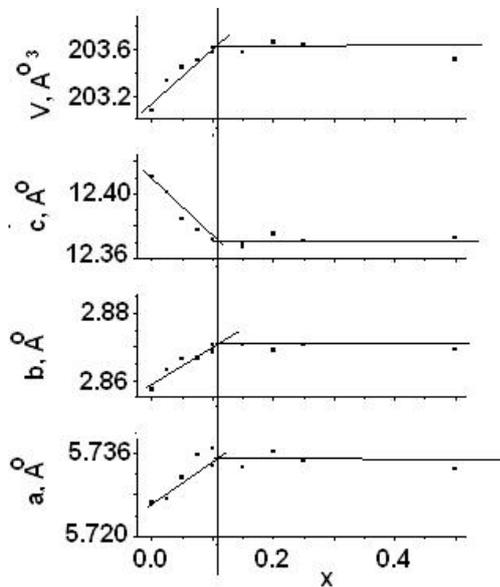
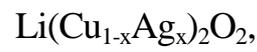
Zn

a, b c

= 0 - 0,12,



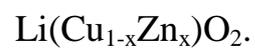
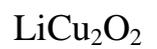
3.12.



3.13.

a, b, c

V



> 0,12

(.3.13).

3.3.4

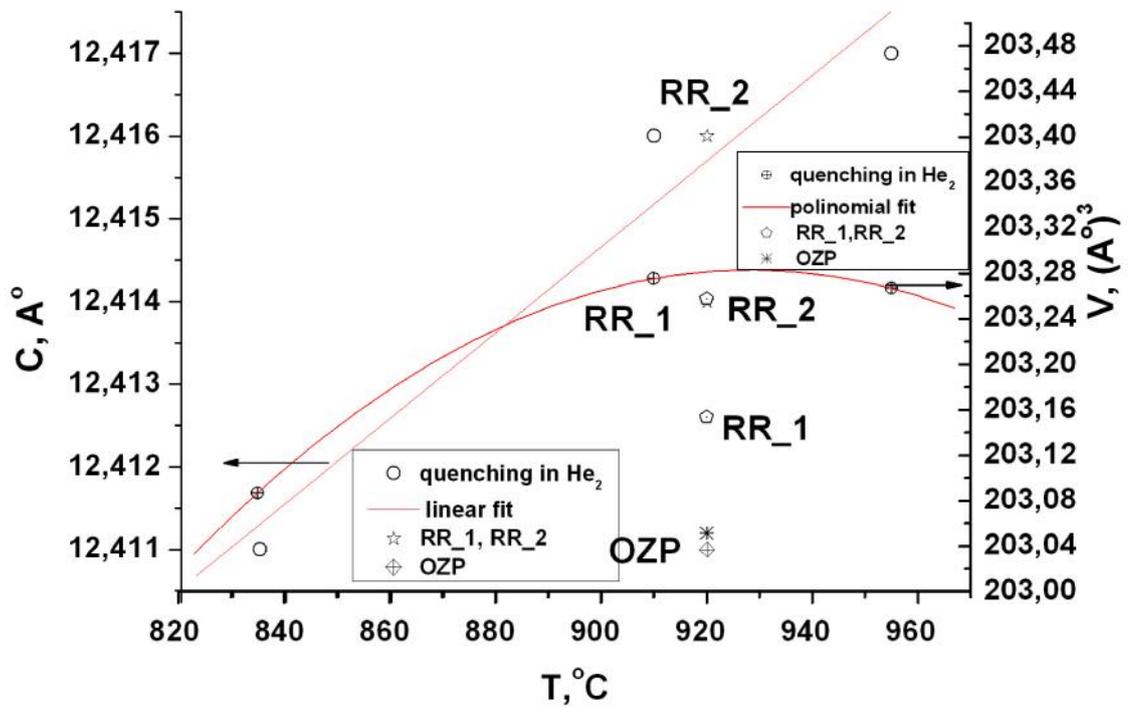
,
 O,
 c (\dots),
 123 [91]) [79],

$p(O_2) = 0,21$, O

. 3.14

(³) c (\dots) ,

:



3.14.

O

- LiCu₂O₂.

		3.5 , (crys_) - (crys) -
A4	12,415(4)	12,412(1) 12,413(5) (crys_1) 12,412(1) (crys_2)
N3	12,418(1)	12,410(1)
W2	12,414(6)	12,412(6)

He₂.

~1% [95].

_1 _2,

~1193 , (

).

c ,

(. .3.1)

. 3.5.

(A4,

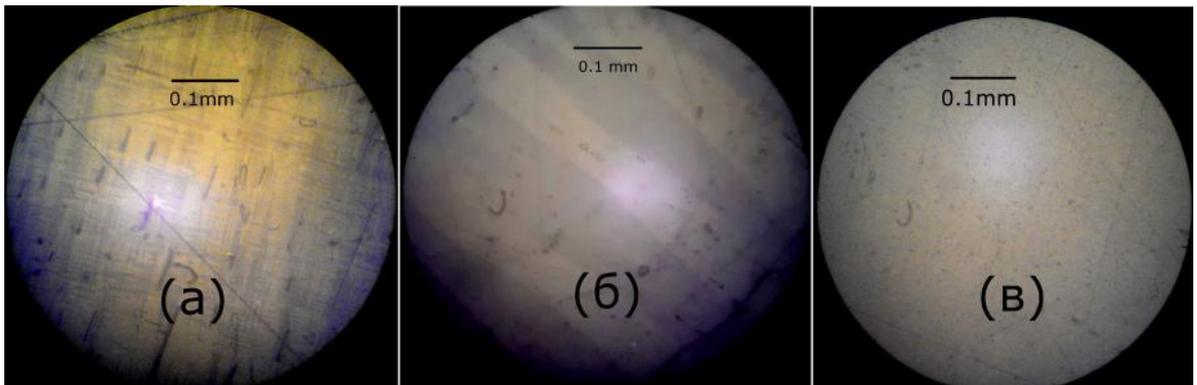
N3),

(001)

LiCu₃O₃,(001) *ab*-

(. 3.15),

[210] [2-10]



3.15.

(a)

()

().

(3.15).

ab

,

,

,

(0012)

2

.

(

).

:

(

.3.3).

W2,

2

,

(001) *ab*-

O

,

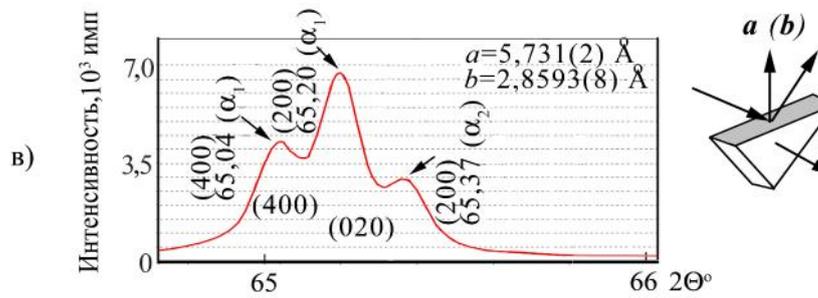
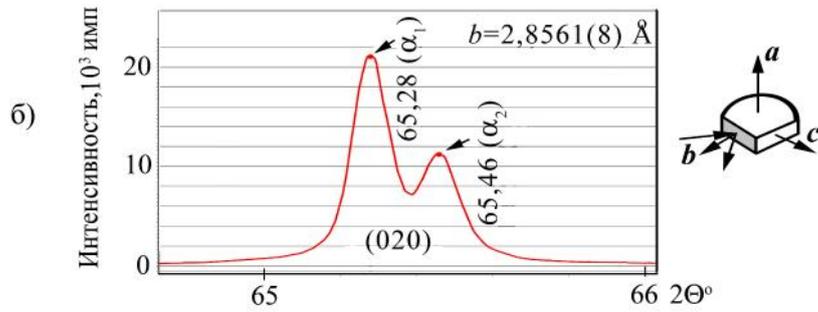
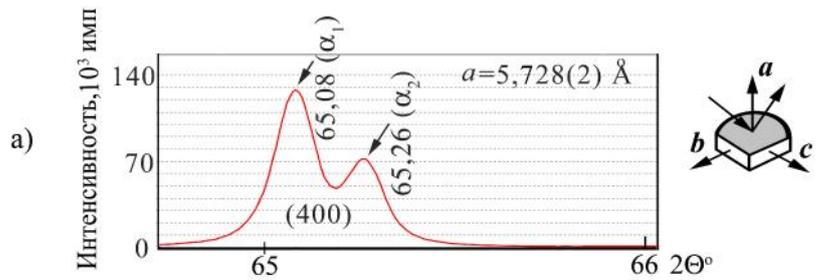
,

.

3.3.5

. 3.16

CuK₁ CuK₂, ((400) – , (020) –) ((400) (020) –).
 1⁻ 2⁻ (400) (020),



3.16.

LiCu₂O₂, ((400) (020) –), –) –).

3.4.



3.4.1



LiCu2O2,

:

(1, 2)

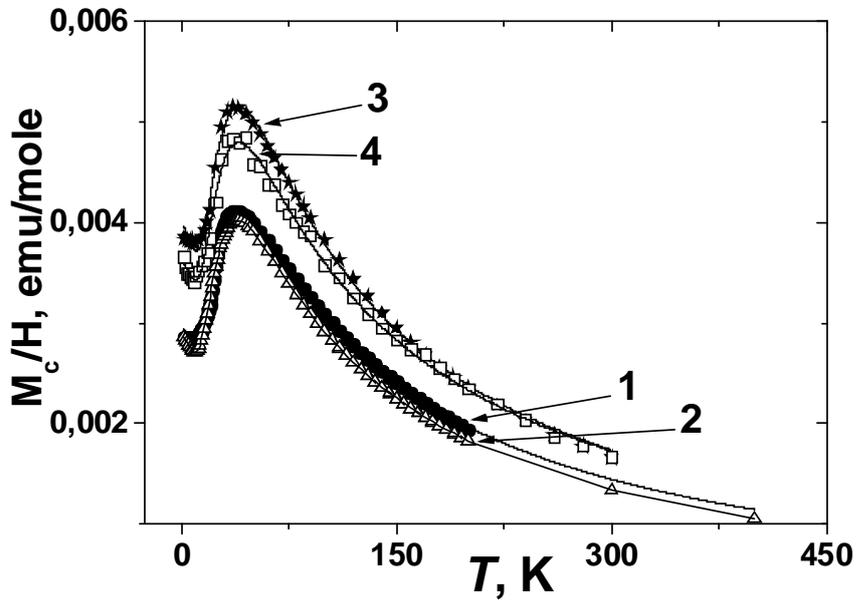
(1, 2).

() = M/H ,

($H = 10$),

40 ,

- (. 3.17).



3.17.

():

|| (1 2)

1

1

$f = 110$

$h_{ac} = 2$ (3, 4).

$\mu_{DC} = 10$ O, () $1 (\dots 3.17).$ $(T) < 400 \text{ K},$ (),

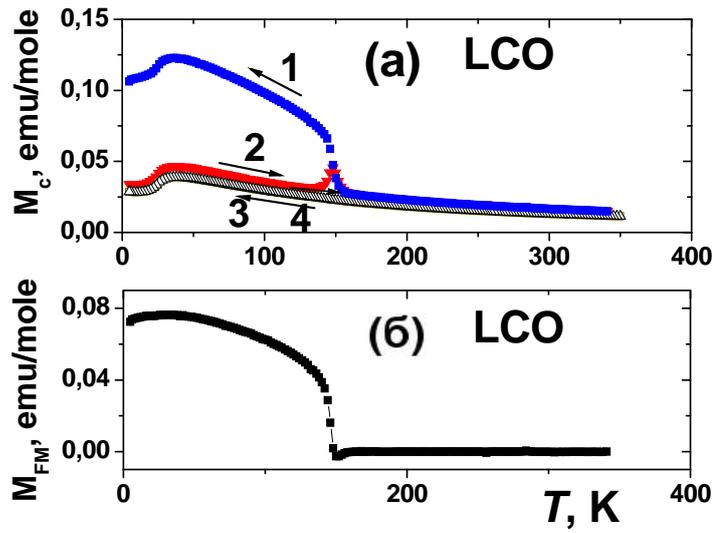
CuO_2 . , , O $\text{CuO}_6,$ uO_4 . u^{2+} - $(\text{uO}_4)^+,$ +1. () . O . , , [92] , La_2CuO_4

$180 - 400$. 3.18 $(T),$ $(H = 10 \dots)$. , (T) $148 \quad 124,7$

$2 \quad 2,$, $(T) \quad H \quad b \quad 2 (\quad 8)$,

$(\text{FC}, \text{ZFC}).$ dM_c/dT $2,$

$N = 24.7$ $\text{LiCu}_2\text{O}_2.$ O $(H = 10 \dots) = 150$



3.19. LiCu_2O_2 ()
 $(H \parallel c, H_{DC} = 10 \text{ Oe})$: (1) FC; (2)
 ZFC, $H \parallel b, H_{DC} = 10 \text{ Oe}$: (3)
 ZFC; (4) FC.)
 $M_{FM} = M_{FC} - M_{ZFC}$ ()
 2.

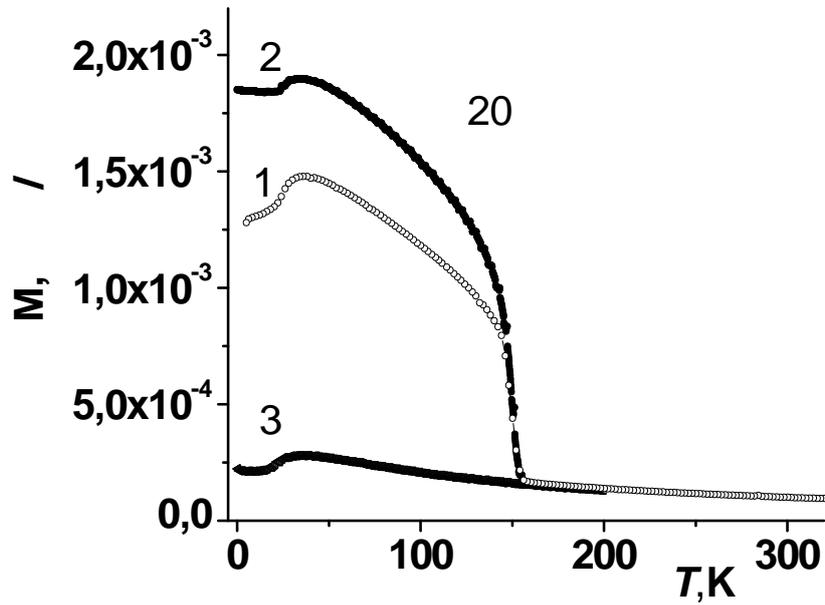
$\text{LiCu}_2\text{O}_{2+x}$
 (. 3.18)

3.4.2

LiCu_2O_2

$\text{Li}_2\text{O} \cdot 4(1-x)\text{CuO} \cdot 4\text{AgNO}_3$ c

0 x 0,5,
 $M(T)$ 5 – 300 (H
 $= 20 \text{ Oe}$), c ZFC (
) FC () ZFC FC $M(T)$
 $T = 37 \text{ K}$ (. 3.20),
 T_{c1} T_{c2}



3.20.

$\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$ $x = 0$ (1), $0,05$ (2) $0,15$ (3)
 20 .

$$dM(T)/dT.$$

$M(T)$ LiCu_2O_2 [68, 73, 94–97].

F $M(T)$ $x = 0,05,$, [69,
 $T_3 = 150$, 85, 90]. T_3 ZFC FC.

Ag ($x > 0,05$)

$T_{c3} = 150$ (. 3.20).

Li^{1+} Cu^{2+} Ag^{2+} .

$\text{LiCu}_2\text{O}_2,$

3.5

LiCu₂O₂

3.5.1

LiCu₂O₂

5 5 2 ³,

, a b

3.21

†(= 0)

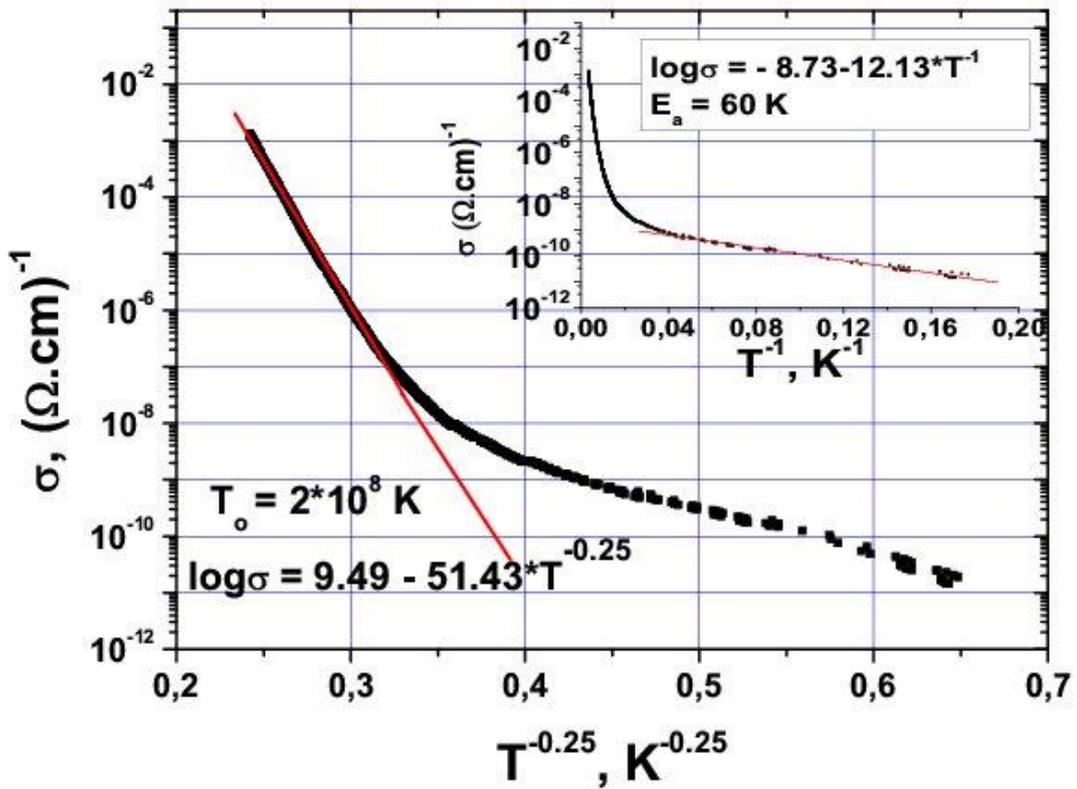
1.

3.21

:

-0.25

T¹ ().



3.21.

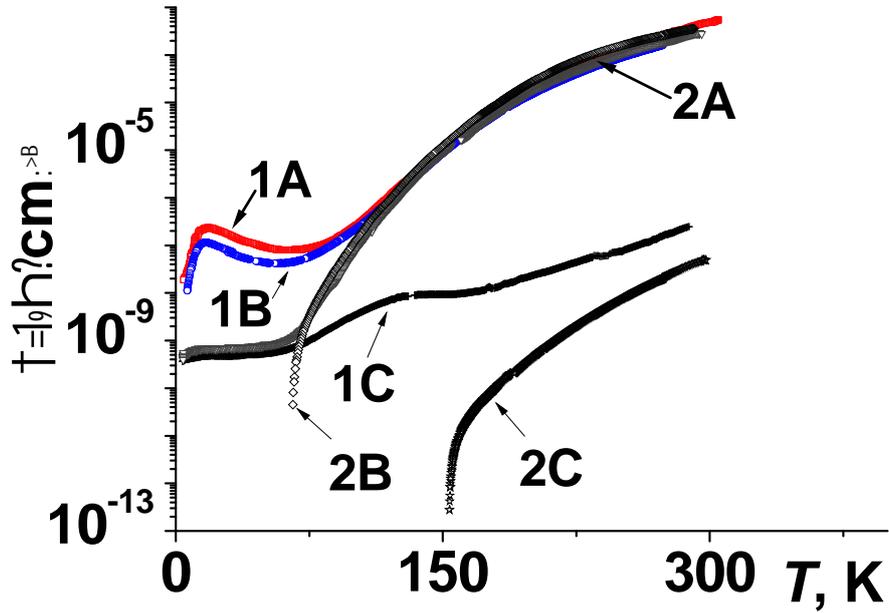
†(0)

LiCu₂O₂

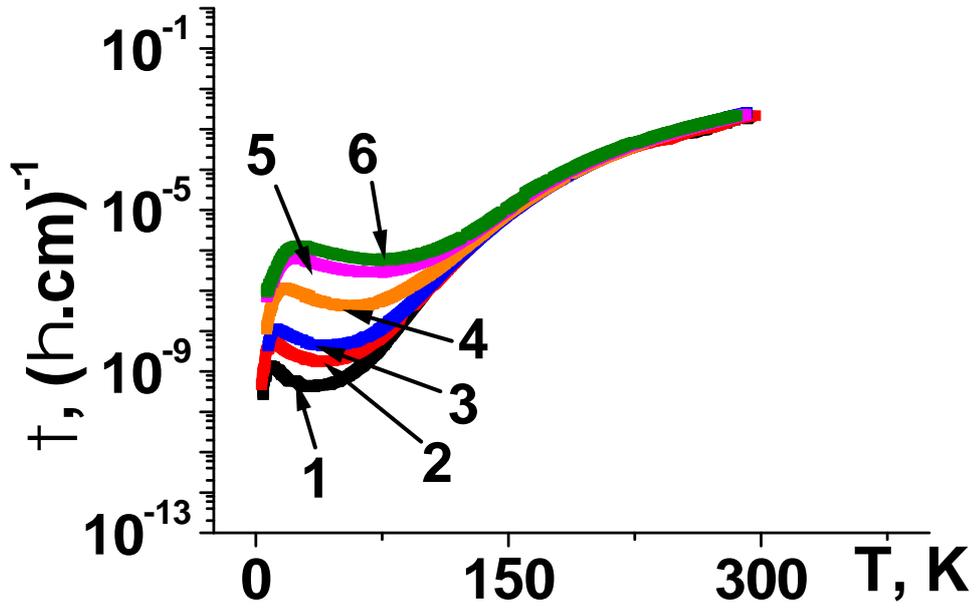
:

-0.25

T¹ ().



3.22. AC DC (1 , 1 1
 10 ~1 B,
 a, b, c; 2 - 4,5 0,3 , 2
 2 - ~10 , M1).
 ,
 .
 300 - 100
 $\rho \sim 10^6 - 10^8$. 300-360
 T^1 $a \sim 0,35 - 0,44$.
 25 DC :
 $a \sim 5 - 6$,
 .
 . 3.22
 AC DC a, b c
 1.
 ,
 [79].



3.23.) (T,) 1 b (1, 2, 3, 4, 5, 6 0,1; 0,5; 1; 10; 50 100 .

DC

T ~ 295 K

ab.

b

a.

$$E_D = -d(\ln \tau) / d(1/T),$$

E_D

0,15; 0,12 0,1

0,3, 107 150 ,

~200

(ln τ ~ 1/T)

(ln τ ~ T^{0.25}).

E_D

b

c

~200

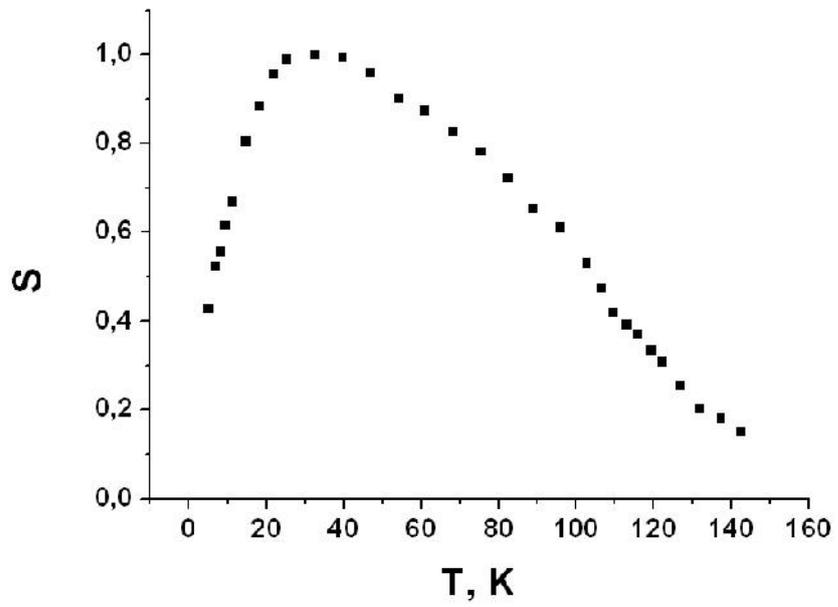
0,158 0,246

1 1

. 3.22.

ab

T <



3.24.

$s()$

1

b.

a b

16.6 15.7 ,

.

c

— ~134 .

()

(1.38).

. 3.23

($T,)$,

M1

b

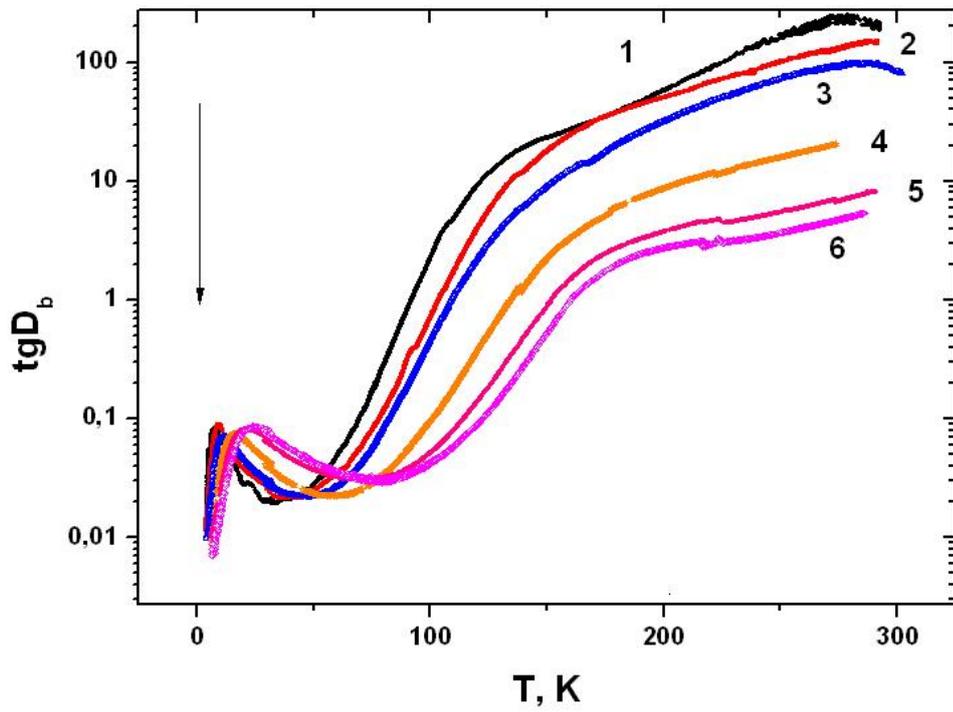
~100

80

80

(1.38): ($T,) = T^n s) s \sim 1,$

()



3.25. $\text{tg}(T)$ b
 0,1; 0,5; 1;10; 50; 100 (1, 2, 3, 4, 5, 6).

(1.38) 0.1–100

()

$s()$ (. 3.24)

[47, 51].

$s(T)$

(1.38) .3.23

4,5 – 80

$s(T)$

.3.24.

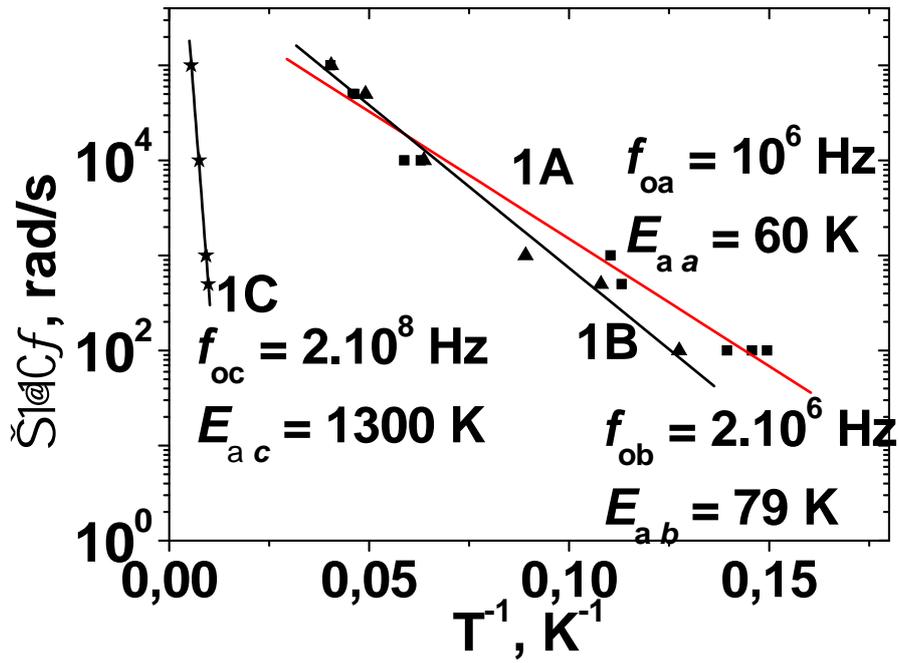
$s(T)$

(

= 30

),

[34, 47, 51, 98].



3.26.

M1

a, b, c.

100 250

()

LiCu₂O₂

tg (T)

4 – 295

F: 0.1 – 100

tg (T) $T = T_{max}$.

. 3.25

tg (T)

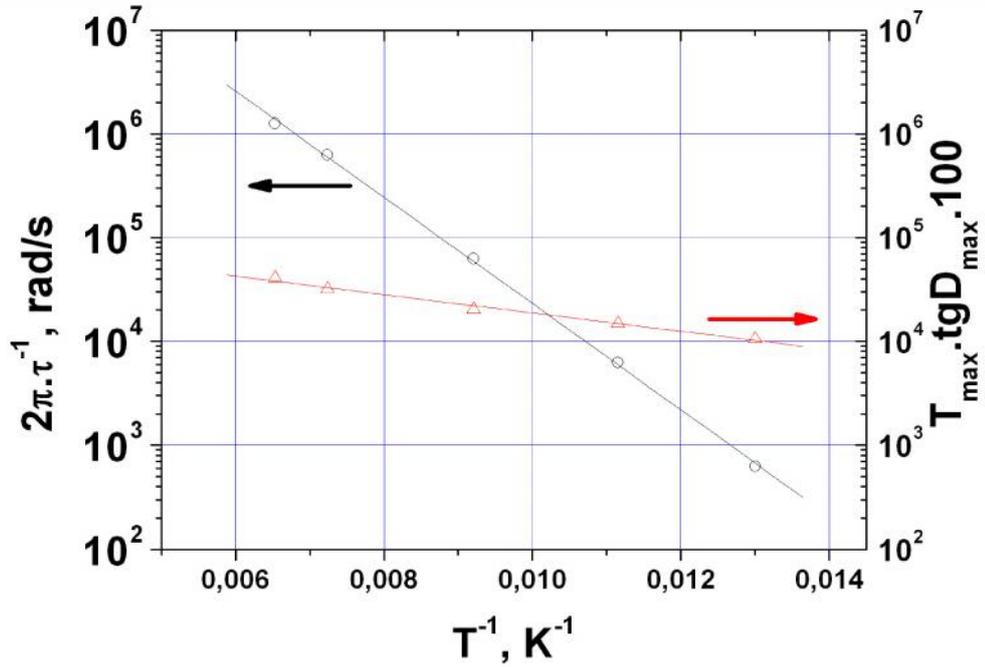
b

(1.39) [47].

tg (T) *a, b, c,*

lgf – $1/T_{max}$,

(. 3.26).



3.27. $f_{max}(T) = (tg_{max} \cdot T_{max})(T)$ M1 **b.**

$$(\omega = 1/2 f),$$

$$= f_{oi} \exp(E_{ai}/kT), \quad i = a, b, c. \quad f_{oi},$$

$\sim 10^6$

$$a \quad b \quad 2 \cdot 10^8 - \dots \quad f_{oa,b} \\ (\dots), \quad E_{a,a,b}$$

CuO₂

$$(b) \quad (\dots) [66]. \quad c$$

$$f_o \quad (\dots).$$

, $< 30 \text{ K}$
, $> 30 \text{ K} -$

[52]:

$$tg_{max}(T^{-1}) \sim [(\dots) \cdot \exp(-Q_d/T)].$$

$$.3.27 \quad 2 / (T^{-1})$$

$$T_{max} \cdot tg_{max}(T^{-1}) \quad E_a, f_o \quad Q_a.$$

(-)
 $E_H = 2Q_a.$

3.5.2

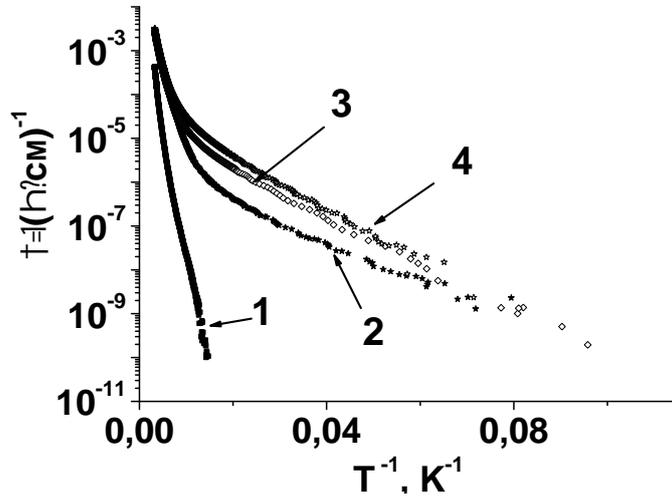
LiCu₂O₂

3.5.2.1

A4.

DC

A4



3.28. DC

A4

c (

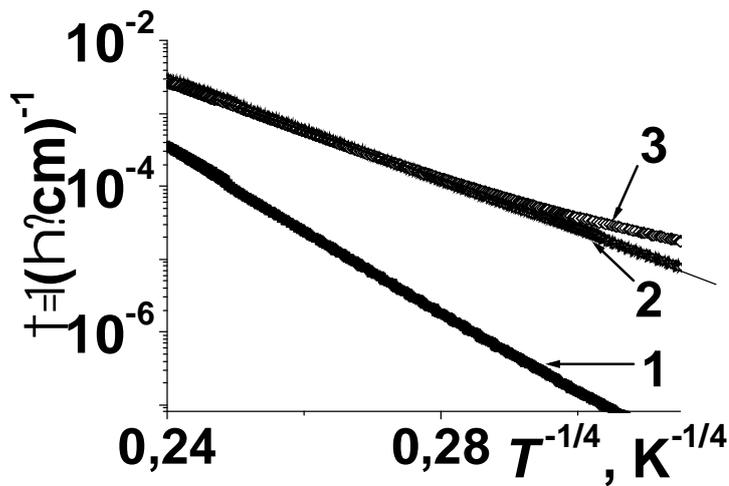
1),

(2)

1113

(3 4)

T^1



3.29. DC

A4 (

)

c (

1),

(2)

1113

(3)

$T^{1/4}$.

ab

20%, *c*

(. 3.28).

1113

40

—

()

c ,

*T*_o

(1.12)

*T*_o.

(~24) —

:

10, 20, 50 100 .

s(T)

s(T)

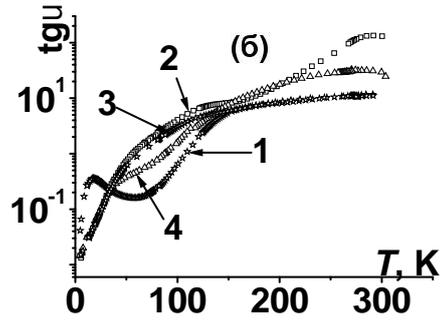
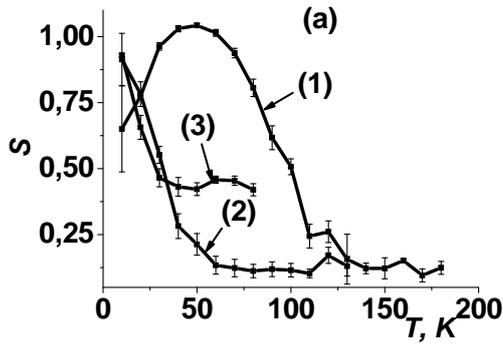
(. 3.30a).

tg (*T*) A4

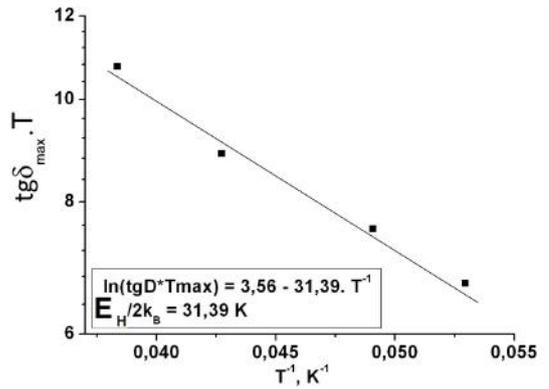
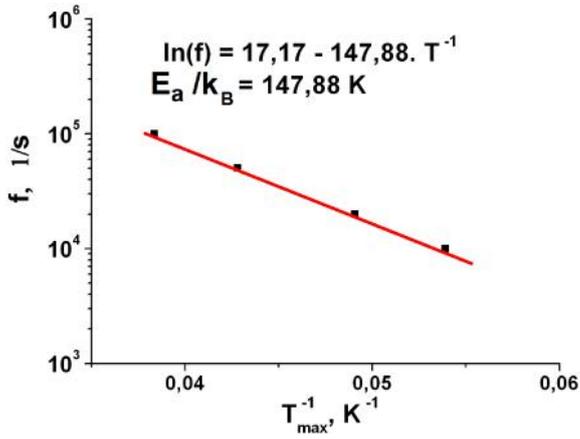
20 , ;

200

(. 3.30).



3.30. a) $s(T)$ ab A4 (1),
 (2) (3);) $tg(T)$ ab
 10 .



3.31. A4 ab
 $f (tg_{max} \cdot T_m)$

$\gg 1$

ab $N = 24$ (
 $a/k_B = 147,9$ $E_H/2k_B = 31,4$ (. 3.31).

() (. 3.30).

3.5.2.2

N3.

2,

2

400

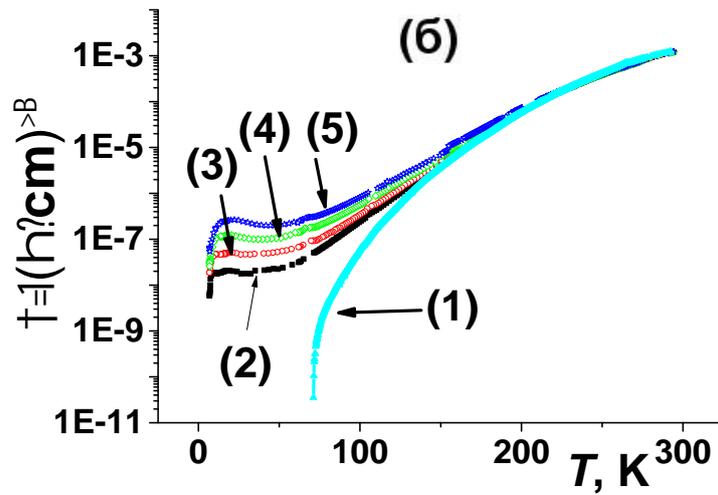
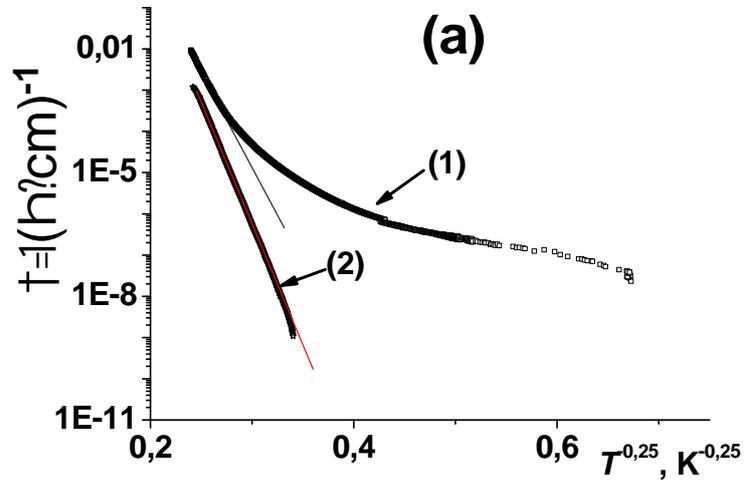
2

1

DC

AC

DC



3.32. a) DC

N3 *ab*

()

(2)

.) DC (

1)

AC (

2,3,4,5

10, 20, 50, 100

)

N3

~ 300 $b = 10^{-2} (\dots)^{-1} \sim a \sim 10^*$ $c,$
 ab $c E_a = 0,16$
 $T_o = 0,97 \cdot 10^8$ ~ 260 $T \sim 230$ $E_a = 0,175$ $T_o =$
 $3,60 \cdot 10^8$ $(\dots 3.32a).$

AC

24 ,

ab,

$(\dots 3.32).$

ab

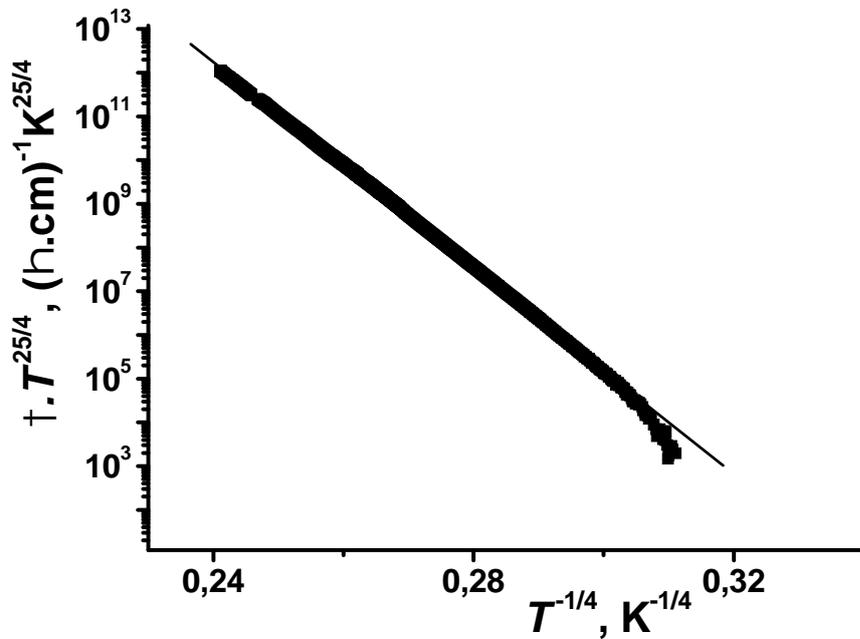
~ 200 .

N3

« »

1.

A4,

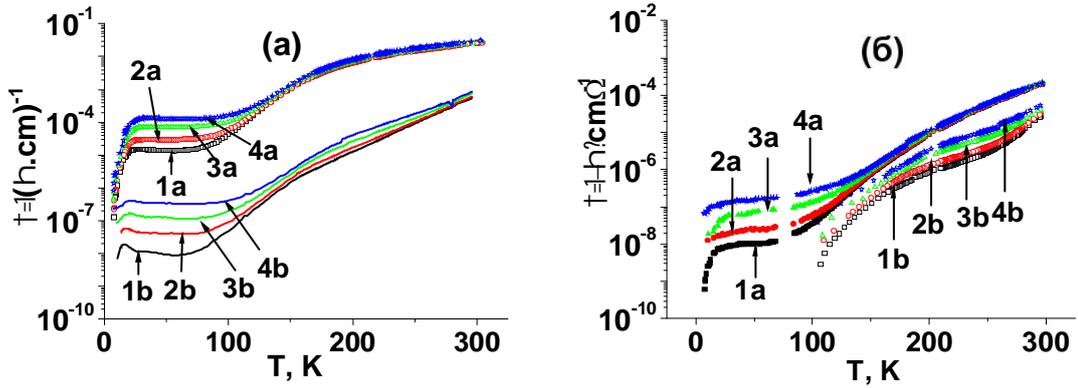


3.33. DC

W2

ab

$\cdot T^{25/4} (T^{-1/4}).$



3.34. AC

W2 () ,

(a) ab () (c a)
 b) 10, 20, 50 100 ()
 1, 2, 3, 4) .

3.5.2.3

W2

W2 T ~ 300

:

$$\tau_b = 2.10^{-2} (\dots)^{-1} \sim \tau_a \sim 10^2 \text{ c.}$$

DC

$$ab \sim 10^{-10} (\dots)^{-1}$$

~50 ,

~100 .

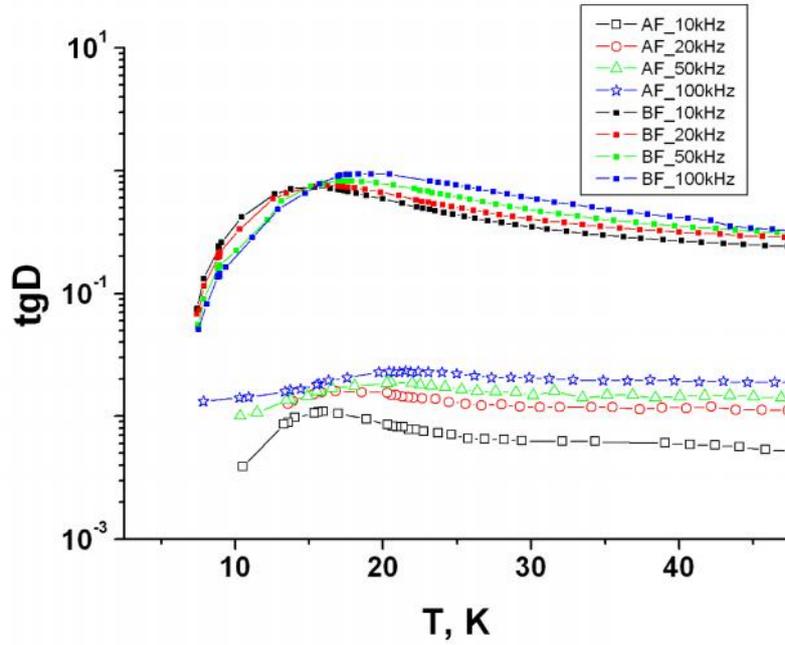
() .

ab

$$m = 25/4 \quad T_o = 6,75.10^7 \quad (\dots \text{ 3.33}).$$

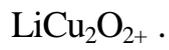
[35]

Li⁺ Cu²⁺



3.35

tg () W2 *ab*



AC

< N

(. 3.34a).

DC

ab,

DC

~300

~100

ab,

AC

tg_{max},

[35, 52, 99],

ab (. 3.35).

ab

3.5.3

LiCu₂O₂

3.5.3.1

Li(Cu_{1-x}Ag_x)₂O₂

4090

Motech

GOM-802

LCR-

(AC)

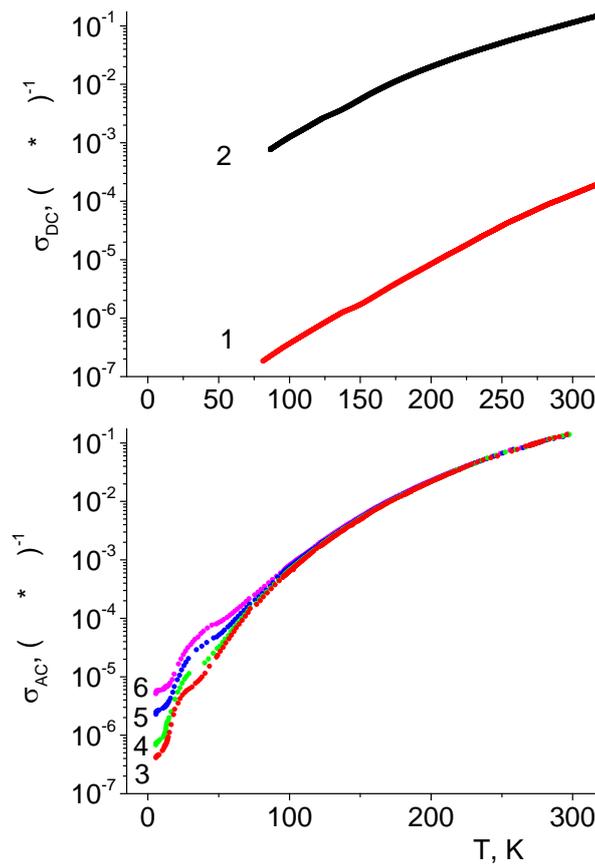
(DC)

[80].

Li(Cu_{1-x}Ag_x)₂O₂

Ag x = 0,15

~3



. 3.36

c x=0 (1) 0,15 (2 - 6),

(1, 2)

10 (

3), 20

(4), 50

(5) 100

(6)

c

DC

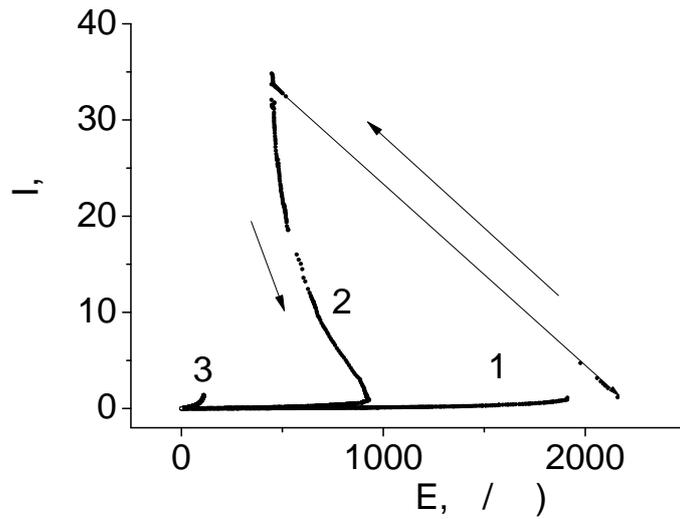
$T < 240$ K

$T \sim 25$ K

$\sigma_{AC}(T)$

(3.36).

Ag ($x = 0,05$),

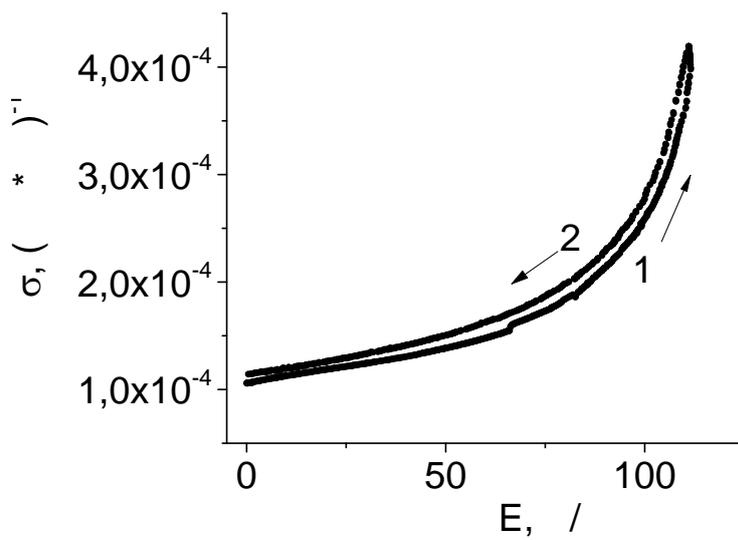


3.37.

$\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$ $x = 0,05$

(1,2) 0,15 (3),
(3) c

(1,2)
78 .



3.38.

$\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$ $x = 0,15$

,
 ab 78 .

S-

(. 3.37),

LiCu₂O₂ [29].

x > 0,05

4

100 / (. 3.38).

3.5.3.2

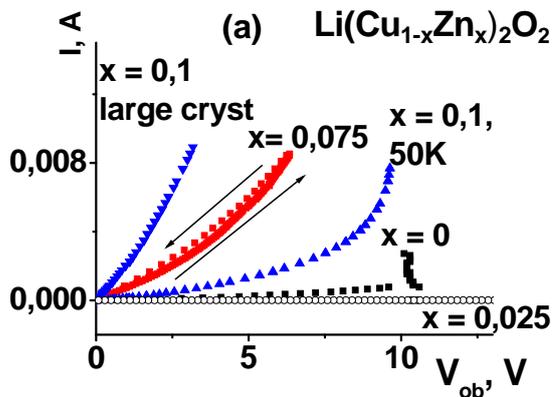
Li(Cu_{1-x}Zn_x)₂O₂

Li(Cu_{1-x}Zn_x)₂O₂

(. 3.39).

Zn

(. 3.39).



3.39. ()

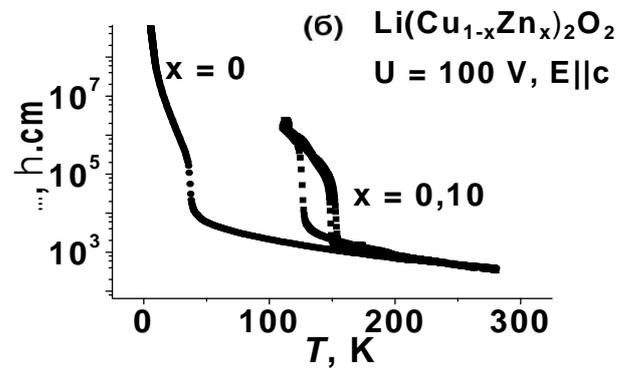
Li(Cu_{1-x}Zn_x)₂O₂

()

(T)

Li(Cu_{1-x}Zn_x)₂O₂,

1



100 .

,

,

.

()

,

.

,

.

1. LiCu₂O₂
Li₂CuO₂-CuO_x
2. LiCu₂O₂
 (Li,Ag)Cu₂O₂, Li(Cu,Zn)₂O₂
3. LCO 12 .% Zn (Cu) 4% Ag (Li).
4. M(T),
 O LiCu₂O₂ (H 10)
 =150 M ||
 O
 Ag (Li_{1-x}Ag_x)Cu₂O₂
 (T < 50)
 x 0,05
 150 ,
5. LCO
 4,2 - 300 0,1 - 10,0
 5 . ~300

(= $\exp(-E_a/kT)$, $E_a= 0,35 - 0,44$)

(= $\exp(-T_0/T^{1/4})$, $T_0=10^6-10^8$ K), 25
 $E = 5 - 6$

5 . LCO,
 $\lg \sim 1/T$ $\lg \sim T^{1/4}$.
 $\therefore a : b : c = 2 : 1 : 10^4$. (295).
 $(T, f) \quad \text{tg}(T, f)$
 $- E_a = 60 - 79$ K,
 $f_r = 10^6$ $E_a = 1300$ K, $f_r = 2 \cdot 10^8$
 ()
 (),

6. Ag Zn LiCu₂O₂
 $x > 0,05$ ~ 3 $x < 0,05$
 $c \quad x > 0,05$

7. LCO
 LCO O O ,

,
1 - 2 ,
,
O ,
.

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